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Optical microspectroscopy study on enriched (11,10) SWCNTs encapsulating C₆₀ fullerene molecules

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1. Introduction

Since the discovery of carbon nanotubes (CNTs) in 1991 by Iijima [1], carbon nanotubes stimulate the scientists' curiosity to explore their unusual properties. Single-walled carbon nanotubes (SWCNTs) exhibit exceptional optical, electronic, and mechanical properties, which can be modified by strain, pressure or doping. Their robust mechanical properties are due to the strong sp² covalent bonds between the carbon atoms network [2].

As far back as more than fifteen years ago, great achievements have been taken in their synthesis, purification, and explanation of their fundamental physical and chemical properties. Important steps have been taken towards realistic and practical applications of SWCNTs, such as nano-electronics, opto-electronics, energy storage devices, flexible electrodes, nano-composites, and sensors [3]. Therefore, several attempts have been made to tailor their electronic properties, such as chemisorption and physisorption of

atoms on the SWCNTs outer surface, substitution of carbon atoms, intercalation of the nanotubes bundles with atoms or molecules, and filling of the nanotubes' inner cavity [4,5]. Among others, filling is attracting great interest. Due to the tubular structure of SWCNTs, they can be filled with various species: metals, metal halides, organic, organometallic compounds, and other substances [5–8]. With this large variety of different fillers, it is possible to fine tune the electronic and transport properties of SWCNTs [7,8].

For example, C₆₀ fullerene molecules can be encapsulated in SWCNTs to form a linear chain of fullerene molecules, so-called peapods [9,10]. Encapsulating fullerene and metallofullerene molecules changes the transport properties of the SWCNTs: Scanning tunnelling spectroscopy studies show that the SWCNTs' band gap is modified at the sites where the C₆₀ fullerene molecules reside [11,12] and the *p*-type SWCNTs field-effect transistor was changed to be ambipolar by encapsulating Gd@C₈₂ metallofullerene molecules [13]. Therefore, it is particularly important to understand the reaction mechanism between the encapsulated fullerene molecules and the SWCNTs. The encapsulation of the C₆₀ fullerene molecules in SWCNTs can be discussed in terms of stabilization energy. For SWCNTs with diameter $d_t < 1.2$ nm the interaction between

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nanotubes and C_{60} molecules is endothermic and the formed peapods system is unstable; whereas the interaction is exothermic for the large SWCNTs diameters $d \approx 1.2\text{--}1.6$ nm and the formed peapods system is stable [14].

Theoretical studies [15,16] predicted that at a critical tube diameter of $d_t \geq 1.37$ nm the unoccupied t_{1u} orbitals of the C_{60} fullerene molecules hybridize with the nearly free electron state (NFE) of the SWCNTs. The NFE state is located between the nanotube and the C_{60} molecule and works as an electron acceptor state [15]. Therefore, the hybridization decreases the electronic density in the vicinity of the SWCNTs, and hence the effective tube diameter of the SWCNTs decreases [15–17].

Optical spectroscopy measurements [18–21] on C_{60} -peapods revealed red and blue shifts of the absorption bands in C_{60} @SWCNTs peapods ($\approx \pm 15$ meV) compared to empty SWCNTs with average tube diameter of 1.37–1.5 nm. Also photoluminescence (PL) measurements [22,23] on C_{60} -peapods over a wide range of diameters ($\approx 1.22\text{--}1.5$ nm) found a change in the band gap of the SWCNTs with the C_{60} fullerene filling, where the band gap modifies in a “ $2n+m$ ” family type-dependent manner: $\Delta E_{11} < 0$ and $\Delta E_{22} > 0$ for type I $\text{mod}[(2n+m),3] = 1$ and $\Delta E_{11} > 0$ and $\Delta E_{22} < 0$ for type II $\text{mod}[(2n+m),3] = 2$. The absorption bands E_{ii} correspond to the i th optical transition in the SWCNTs. For SWCNTs with average tube diameter of 1.37–1.5 nm, the maximum change of the band gap was around $\approx \pm 20$ meV.

All the previous SWCNTs samples used for the preparation of the C_{60} -peapods were of multi-chirality character, which prevented a detailed understanding of the interaction between the nanotubes and the fullerene molecules. Single chirality SWCNTs have unique electronic properties and diameter. Therefore, they are ideal candidates for the encapsulation of C_{60} molecules, in order to understand the interaction mechanism between the nanotube and the fullerene molecules. This would potentially open the door to control the chemical and physical properties of SWCNTs, in order to implement such nanostructures in electronic devices.

Here, we report on the preparation of high-yield enriched (11,10) SWCNTs encapsulating C_{60} fullerene molecules, so called (11,10)-peapods, and also the preparation of DWCNTs derived from the (11,10)-peapods. Raman spectroscopy is used to prove the formation of inner wall inside the (11,10) SWCNT. Optical microspectroscopy is applied to study the effect of the interaction between the fillers (C_{60} fullerene molecules, inner tube) and the (11,10) SWCNT on the electronic properties.

In general, optical and infrared microspectroscopy is a powerful technique to characterize the electronic band structure in terms of the energy position and spectral weight of the excited interband and intraband transitions. As demonstrated recently, the optical response is capable of monitoring small changes in the electronic band structure of SWCNTs.

2. Experimental

The single chirality (11,10) SWCNTs (1.44 nm diameter) were prepared using two purification approaches with Cesium chloride (CsCl) sorting and metal-semiconducting (MS) sorting by usual density-gradient ultracentrifugation. The detailed preparation method is described in Ref. [24]. The fraction containing the (11,10) SWCNTs was extracted carefully from the top of the centrifugation tube. Vacuum filtration method was used to prepare bucky-paper from the (11,10) SWCNTs over cellulose nitrate substrate. A small piece of the (11,10) SWCNTs film was transferred in an acetone bath. The free standing film was fished by using a stainless steel substrate with many holes in the middle (see Fig. 1 a–c for illustration).

Bundled multi-chirality SWCNTs were purchased from Carbon Solutions Inc. (Type P2, average diameter 1.4 nm). The P2 carbon

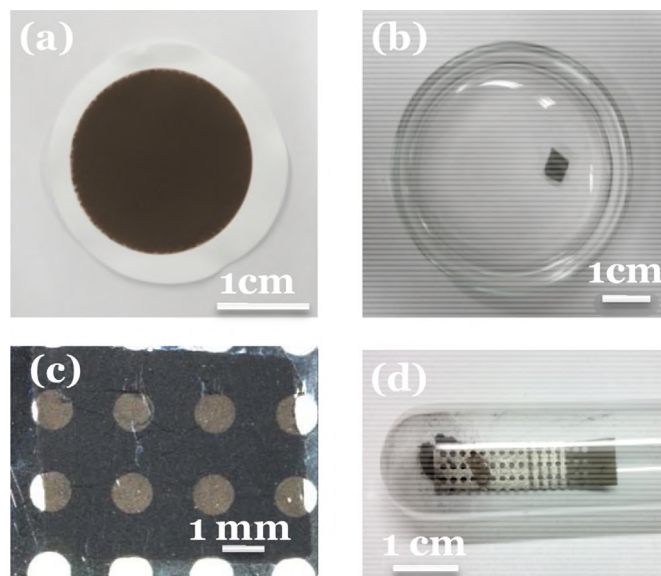


Fig. 1. (a) Photograph of the (11,10) SWCNT bucky paper. (b) Free-standing carbon nanotube film floating in acetone solution after dissolution of the cellulose nitrate membrane in the acetone solution. (c) Optical image of the free-standing film over stainless steel substrate. (d) Optical image of the closed evacuated quartz tube containing the (11,10) SWCNT free-standing film over stainless steel substrate and C_{60} powder. (A colour version of this figure can be viewed online.)

nanotubes were prepared using the arc discharge method. The received P2 carbon nanotubes sample was purified by the supplier to $>90\%$ carbon content purity. The sample was used without any further purification process. Vacuum filtration was used to prepare bucky-paper multi-chirality SWCNTs over cellulose nitrate substrate from Triton X-100 suspension. The above described procedure (shown in Fig. 1 a–c) has been applied to prepare free-standing films of P2 SWCNTs for optical spectroscopy measurements.

The C_{60} peapods were prepared by the gas phase method using the (11,10) SWCNT film prepared as described above [19,25]. The typical preparation method is as follows: the caps of the SWCNTs were opened by heating the SWCNTs film over the stainless steel substrate in a furnace at 575°C for 30 min. This process is important to ensure that all the caps are opened and to remove any residuals of amorphous carbon that could be attached to the surface of the SWCNTs. Weighted amount of C_{60} fullerene powder and the (11,10) SWCNT film were degassed under dynamic vacuum at 200°C for 24 h; after that both samples were sealed in quartz tube under vacuum and heated at 750°C for 5 days continuously (see Fig. 1 d). To remove the non-reacted fullerene molecules from the surface of the (11,10) SWCNTs, the sample was heated at 700°C under dynamic vacuum for 2 h [26,27]. The filling of the (11,10) SWCNT film with C_{60} molecules enables to measure the change of the optical properties due to the fullerene filling in a quantitative way. Based on our previous work [19,28] and literature data [29,30] on C_{60} peapods prepared by the gas phase method we can estimate the filling ratio to be higher than 95%.

Transmission data were collected at ambient conditions in the frequency range $2500\text{--}23,000\text{ cm}^{-1}$ using a Bruker IFS 66v/S Fourier transform infrared spectrometer in combination with an infrared microscope (Bruker IR Scope II) with a $15\times$ magnification objective. The intensity $I_{\text{sample}}(\omega)$ of the radiation transmitted through the free-standing film and the intensity $I_{\text{ref}}(\omega)$ of the radiation transmitted through air, as reference, were measured. From $I_{\text{sample}}(\omega)$ and $I_{\text{ref}}(\omega)$ the transmittance and absorbance spectra were

calculated according to $T(\omega) = I_{\text{sample}}(\omega)/I_{\text{ref}}(\omega)$ and $A(\omega) = -\log_{10} T(\omega)$, respectively. Room-temperature Raman spectra were measured with a 515 nm excitation line of an argon laser and recorded by a triple Raman spectrometer T64000 (Horiba Jobin Yvon), interfaced to an Olympus BX-40 microscope (100 \times LWD objective). The laser power impinging on the sample was 0.15 mW. The Raman spectrometer was calibrated by using the mercury line at 435.8 nm.

3. Results and discussion

The background-subtracted absorbance spectra of (a) (11,10) SWCNTs and (b) multi-chirality SWCNTs free-standing films are depicted in Fig. 2. The absorption bands S_{ii} and M_{ii} correspond to the i th optical transition in the semiconducting and metallic SWCNTs, respectively. The inset of Fig. 2 depicts the raw absorbance spectrum of the free-standing (11,10) SWCNTs film and the absorbance spectrum after the background subtraction. The broad background is due to the π - π^* absorption centered at around 5 eV and follows an approximately linear behavior. The absorption bands were fitted with Lorentzian oscillators and the contributions are listed in Table 1. In general, the spectrum of the (11,10) SWCNTs is characterized by three absorptions bands at ≈ 0.69 , 1.25, and 2.2–2.4 eV. The S_{11} , S_{22} , and $S_{33/44}$ bands are due to the first, second, and higher optical transitions in the semiconducting tubes. The absence of significant metallic contributions M_{11} in the energy range around ≈ 1.7 eV signals the semiconducting character of the

Table 1

Peak positions in cm^{-1} of the Lorentz contributions of the S_{11} , S_{22} , M_{11} and $S_{33/44}$ for SWCNTs, peapods, and DWCNTs.

	S_{11}	S_{22}	M_{11}	$S_{33/44}$
Multi-chirality SWCNTs	4800 5255 6920	8960 9610 10,330	13,300 14,180 15,098	17,763 18,873 19,909
(11,10) SWCNTs	5575	10,050		16,996 18,050 18,950
Multi-chirality Peapods	4866 5350 6940	9020 9676 10,340	13,440 14,260 15,120	16,875 17,911 18,947
(11,10) Peapods	5114	9676		20,257 17,474 18,405
(11,10) DWCNTs	5327	9720		17,584 18,460

sample. The S_{11} and S_{22} absorption bands can be described by one strong Lorentzian oscillator respectively, which reflects the single-chirality nature of the sample. The small Lorentzian contributions close to the S_{22} absorption bands indicate a small amount of other chiralities besides (11,10) SWCNTs.

In comparison, the background-subtracted absorbance spectrum of the multi-chirality SWCNTs free-standing film is shown in Fig. 2 (b). The fine structure of the S_{11} , S_{22} , M_{11} , and $S_{33/44}$ bands reflects the rather large diameter and chirality distribution in the sample. The absorption bands were fitted with Lorentzian oscillators and the contributions are listed in Table 1. Our previous work [19,31] showed that the P2-SWCNTs have a diameter distribution from 1.3 to 1.5 nm with an average tube diameter of 1.42 nm.

It is clear from Table 1 that the S_{11} , S_{22} , and $S_{33/44}$ absorption bands in the multi-chirality SWCNTs free-standing film are redshifted compared to the (11,10) SWCNTs. This red shift could be attributed to the bundling effect, as explained in the following. The multi-chirality SWCNTs free-standing film was prepared by sonicating the P2-SWCNTs powder in a Triton X-100 solution, then the solution was left over night to allow the large bundles to settle down. The clear supernatant solution was filtered over a cellulose nitrate film. However, in the case of the (11,10) SWCNTs, before the chirality separation the SWCNTs powder was dispersed into a 2% (w/w) sodium cholate (SC) and 42.5% (w/w) CsCl solution by using tip homogenizer for 4 h, then the solution was centrifuged for 1 h at 40,000 rpm. The SWCNTs supernatant was used for the single-chirality separation. It has been shown that the sonication and centrifugation of bundled SWCNTs decrease the bundling to a limiting value associated with individually dispersed SWCNTs [32]. This means that the degree of bundling in the case of the multi-chirality SWCNTs free-standing film is higher than that in the case of the (11,10) SWCNTs free-standing film. Previous theoretical and experimental studies [33,34] showed that the bundling effect leads to a lowering of the symmetry of individual SWCNTs in a bundle. Therefore, the separation between the valence and the conduction band is reduced, resulting in a red shift of tens of meV in the optical absorption bands.

To clarify the effect of the C_{60} filling on the optical properties of the (11,10) SWCNTs, we plot in Fig. 3 the background-subtracted absorbance spectrum of (a) (11,10) peapods and (b) multi-chirality peapods free-standing films. The absorption bands were fitted with Lorentzian oscillators, and the contributions are listed in Table 1.

Compared to the empty (11,10) SWCNTs [Fig. 2 (a)] one observes that all absorption bands in the (11,10) peapods are redshifted (≈ 60 meV for S_{11} , 50 meV for S_{22} , and 67–70 meV for $S_{33/44}$), broadened, and lost spectral weight. See also Fig. 5 for a direct

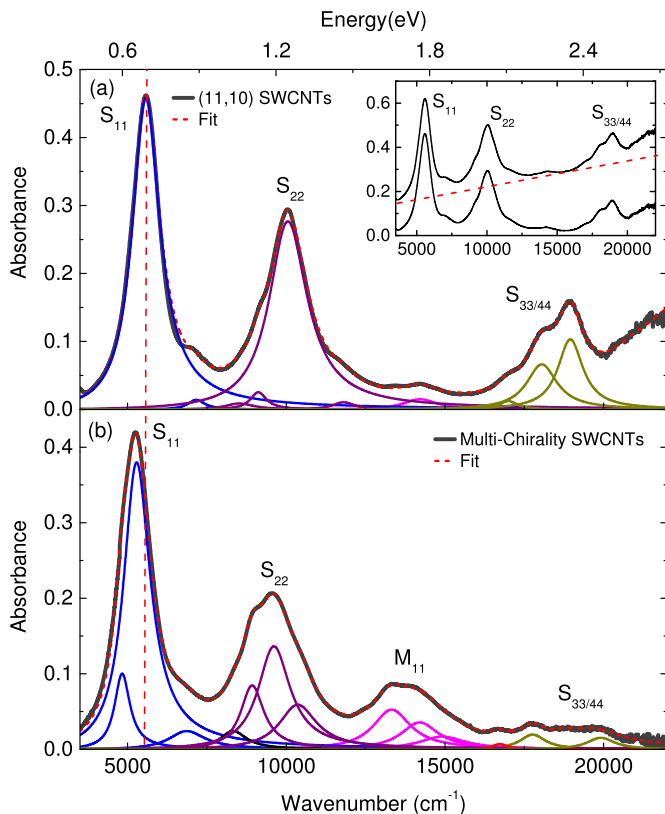


Fig. 2. Background-subtracted absorbance spectra of (a) (11,10) SWCNTs and (b) multi-chirality SWCNTs together with the fit of the absorption bands using Lorentzian oscillators. The absorption bands S_{ii} and M_{ii} correspond to the i th optical transition in the semiconducting and metallic SWCNTs, respectively. The vertical, dashed line marks the position of the S_{11} band in (11,10) SWCNTs. Inset: absorbance spectrum for the (11,10) SWCNTs film together with the linear background (dashed red line) and the absorbance spectrum after background subtraction. (A colour version of this figure can be viewed online.)

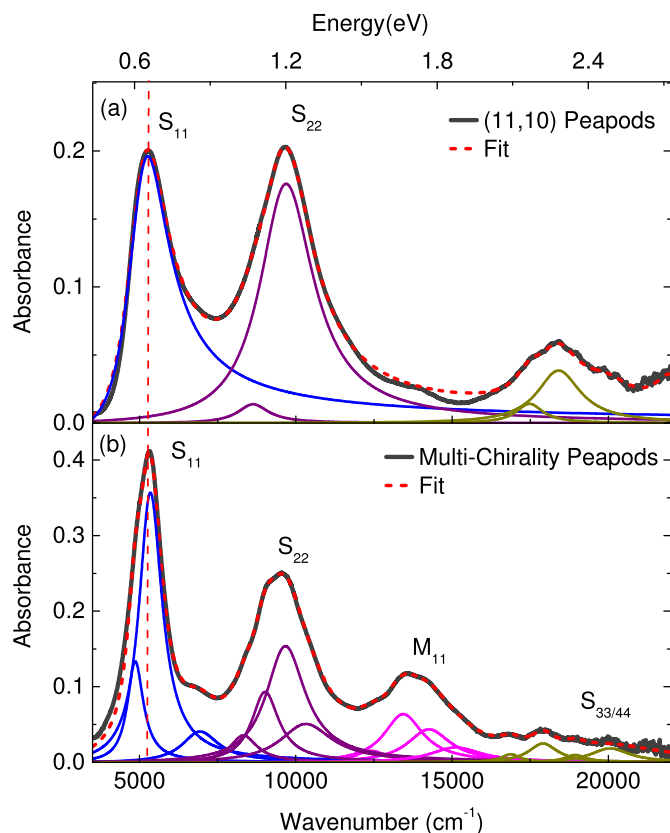


Fig. 3. Background-subtracted absorbance spectra of (a) (11,10) peapods and (b) multi-chirality peapods together with the fit of the absorption bands using Lorentzian oscillators. The absorption bands S_{ii} and M_{ii} correspond to the i th optical transition in the semiconducting and metallic SWCNTs, respectively. The vertical, dashed line marks the position of the S_{11} band in (11,10) peapods. (A colour version of this figure can be viewed online.)

comparison between the results for (11,10) SWCNTs and (11,10) peapods. When compared to literature data, this behavior is not expected for our peapods system. For the multi-chirality peapods, no obvious changes in the spectrum in comparison to multi-chirality SWCNTs [Fig. 2 (b)] are found, except for the slight blue shift in the energy position of the absorption bands [see Table 1]. According to our previous work [19] on multi-chirality peapods, the blue shift of the absorption bands can be attributed to the hybridization between the NFE state of the SWCNTs and the t_{1u} state of the C_{60} , since hybridization is expected to reduce the effective tube diameter [15,16], and hence the optical absorption bands are blue-shifted according to the Kataura plot [35].

Now we will discuss the optical properties of (11,10) peapods as compared to empty (11,10) SWCNTs. In general, the redshift, broadening, and loss of spectral weight of the optical absorption bands could be attributed to strain-induced effects, the hybridization between the t_{1u} orbitals of the C_{60} molecules and the NFE states of the SWCNTs [22,23] or the symmetry breaking of the nanotubes [28,31,36,37]. The band gap modification upon insertion of C_{60} molecules was found to be diameter- and chirality-dependent [22,23]. The band gap related to the first and the second optical transitions, E_{11} and E_{22} , respectively, even changes in opposite direction between type I and type II SWCNTs due to strain-induced effects [22,23]. For example, strain/hybridization leads to a redshift of 66 meV for E_{22} and a blue shift of 12 meV for E_{11} in (13,5) SWCNTs (type I), whereas for the (15,2) SWCNTs (type II) a redshift of 60 meV for E_{11} and a blue shift of 32 meV for E_{22} were found. For

larger diameters ($d_t \geq 1.3$ nm), the shifts in energy position of the optical transitions E_{11} and E_{22} approach zero and change their sign [22,23]. A photoluminescence study by Okubo et al. [22] showed that the induced changes in the S_{11} and S_{22} transitions is zero in (11,10) SWCNTs encapsulating C_{60} fullerene molecules. Therefore, hybridization between the electronic states in (11,10) SWCNTs and the C_{60} molecules cannot be considered as the primary reason for the observed large redshift of the optical absorption bands. Neither can they be related to strain-induced effects, since theoretical calculations [15,16] predict that these do not occur for diameters $d_t \geq 1.3$ nm (the diameter of (11,10) SWCNTs is 1.44 nm).

A theoretical study on the energetic and the mechanical stability of C_{60} peapods using density-functional theory (DFT), molecular mechanics, and molecular dynamics simulations [38,39] showed that the interaction between the C_{60} fullerene molecules and the nanotube wall strongly depends on the space between them. For (10,10) SWCNTs the inter-space amounts to 3.325 Å, which is very close to the graphite interlayer space (3.35 Å) [40]. Therefore, homogeneous attractive van der Waals (vdW) interactions between the C_{60} fullerene molecules and the nanotube wall from both sides are to be expected, leading to a homogeneous arrangement of the C_{60} molecules in the center of the nanotube. As the nanotube diameter increases, the inter-space between C_{60} molecules and the nanotube wall increases and the vdW attraction will be one-sided. Meaning that the encapsulation of the fullerene molecules will enhance the deformation of the nanotubes for larger diameter than the (10,10) SWCNTs [38]. In the case of (11,10) peapods, the inter-space between the SWCNTs and the C_{60} molecules amounts to 3.67 Å, which is, however, still close to the graphite interlayer space. Therefore, the redshift and loss of spectral weight of the absorption bands cannot be attributed due to the non-homogenous interactions between the C_{60} molecules and the nanotube wall.

It is important to note that the optical excitations in carbon nanotubes cannot be explained in terms of simple single electron interband transitions. The optical excitations are found to be of excitonic nature [30], and hence their energetic position is a combination of the band gap energy and the exciton binding energy [30,41]. It has been furthermore shown that the dielectric environment affects the optical excitation energies in carbon nanotubes [42–47].

As the dielectric constant increases, the Coulomb interactions (electron-hole) and the self-energy (electron-electron repulsion) are screened. The decrease of the self-energy always exceeds the exciton binding energy, which results in an overall redshift of the optical transitions [46–48]. The redshift is consistent with the theoretical calculations [47–51] and experimental results [46,52]. A maximum redshift of 55 ± 4 meV for E_{11} and 50 ± 10 meV for E_{22} was observed. The redshifts saturate at an environmental dielectric constant of ~ 5 , without any significant (n,m) dependence [46,48,52].

Furthermore, theoretical calculations by Miyauchi et al. [47] demonstrated that the screening effect is related to the static dielectric constant ϵ as a linear combination of both the dielectric constant of the surrounding environment ϵ_{env} and the internal environment, which is called the dielectric constant of the nanotube itself, ϵ_{tube} . Hence, the dielectric constant ϵ is calculated according to the equation:

$$\frac{1}{\epsilon} = \frac{C_{env}}{\epsilon_{env}} + \frac{C_{tube}}{\epsilon_{tube}} \quad (1)$$

where C_{env} and C_{tube} are coefficients for the outside and inside of a nanotube, respectively [47]. Here, the nanotube and its surrounding medium are considered as a series connection of two capacitors with different dielectric constants where the electric flux lines pass

through the two dielectric materials. The constants C_{env} and C_{tube} are dimensional constants, i.e., their values depend on the nanotube diameter [47], therefore both values will not change after the filling. In our studies the optical properties of the (11,10) SWCNTs and (11,10) peapods free-standing films were measured in air, i.e., $\epsilon_{env} \approx 1$. Accordingly, the dielectric screening will depend only on the internal dielectric constant of the nanotube ϵ_{tube} , which is affected by the filling. Assuming the dielectric constant of the encapsulated C_{60} fullerene molecules to be 4–4.5 [53,54], we attribute the observed redshift of the optical absorption bands in the case of the (11,10) peapods as compared to (11,10) SWCNTs to an increase of the internal dielectric constant ϵ_{tube} of the nanotubes after filling.

The decrease in spectral weight of the S_{11} and S_{22} absorption bands in (11,10) peapods compared to empty SWCNTs could be attributed to the charge transfer from the nanotubes to the fullerene molecules. Theoretical [55] and experimental [20] studies showed that electrons are transferred from SWCNTs to C_{60} molecules. Therefore, C_{60} molecules are considered as a hole dopant [20,55–57]. It has been shown that [58,59] the p-doping reduces the electron density in the semiconducting SWCNTs and lowers the Fermi level into the valence band toward the first van Hove singularity or below, and hence the S_{11} and S_{22} absorption bands are partially quenched [60].

In order to test the above stated assumptions regarding the effect of the internal dielectric constant and the charge transfer between the nanotubes and the C_{60} molecules, the same free-standing (11,10) peapods film was transformed to DWCNTs using the laser irradiation method [61,62]. Here, we aim to change both the internal dielectric constant and the charge transfer between the nanotubes and the C_{60} molecules. The sample was irradiated with 514 nm laser (250 mW) for 5 min under high vacuum. The activation barrier for the C_{60} molecules polymerization was found to be 1.4 eV [38] indicating that the 514 nm (2.4 eV) is suitable to overcome the activation barrier. We also tried the thermal transformation of the (11,10) peapods to DWCNTs under dynamic vacuum (at 1200 °C for 24 h), but it was not possible in this case. As shown in Fig. 1 (d) the free-standing film is very small (1 mm²) and fragile, therefore the dynamic vacuum with the very high temperature destroyed the nanotube film. After laser irradiation a small piece of the (11,10) DWCNTs free-standing film and of the (11,10) SWCNT film were transferred on a SiO_2 substrate for Raman measurements.

The inset of Fig. 4 depicts the Raman spectrum of the empty (11,10) SWCNTs, before filling, as a reference. The spectrum was fitted with Lorentzian functions, in order to specify the various contributions. The obtained peak positions are given in the spectrum. Accordingly, we observe the radial breathing mode (RBM) of the outer tubes at 173 cm⁻¹. The small shoulder at 183.3 cm⁻¹ could be attributed to a different chirality present in the (11,10) SWCNTs sample. The diameter d of the (11,10) SWCNTs was calculated according to the relation $\omega_{RBM} = 248/d$ and amounts to $d \approx 1.43$ nm (This formula is used for carbon nanotubes dispersed on SiO_2 substrate) [63]. The obtained value matches that of the diameter of the (11,10) SWCNTs [24].

Fig. 4 shows the Raman spectrum of the (11,10) DWCNTs including the fit with Lorentzian functions. For the (11,10) DWCNTs, the RBM of the outer tube is observed at 166.6 cm⁻¹. The small shoulder observed at 176.3 cm⁻¹ is attributed to a different chirality present in the starting (11,10) SWCNTs sample. At higher frequencies, one can observe a peak at 363.3 cm⁻¹ with quite low intensity in addition to some weak contributions close to this peak. In contrast, in the case of (11,10) SWCNTs (see the inset of Fig. 4) no Raman features are observed for the empty tubes. In general the Raman modes detected in this frequency range (300–400 cm⁻¹) for

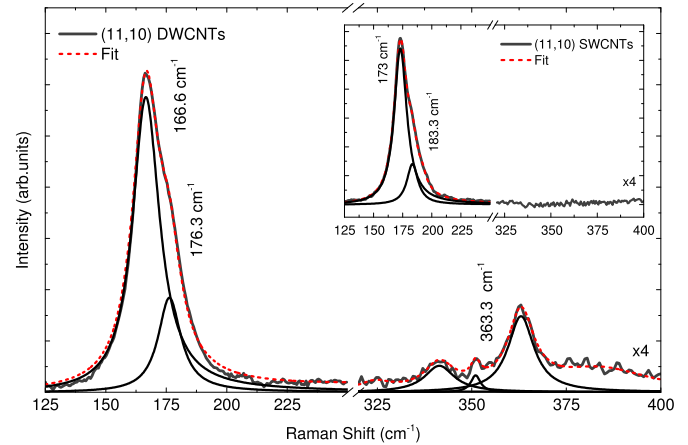


Fig. 4. Raman spectrum recorded with a 515 nm excitation wavelength showing the RBMs for (11,10) DWCNTs. Modes in the range 300–400 cm⁻¹ correspond to the RBMs of the inner tubes. Raman spectra in the range 300–400 cm⁻¹ were multiplied by a factor of 4. The dashed-red line is a fit to the data with Lorentzian functions. The Lorentz contributions are shown as well. Inset: Raman spectrum recorded with 532 nm excitation wavelength showing the RBMs of the empty (11,10) SWCNTs together with a fit with Lorentzian functions. (A colour version of this figure can be viewed online.)

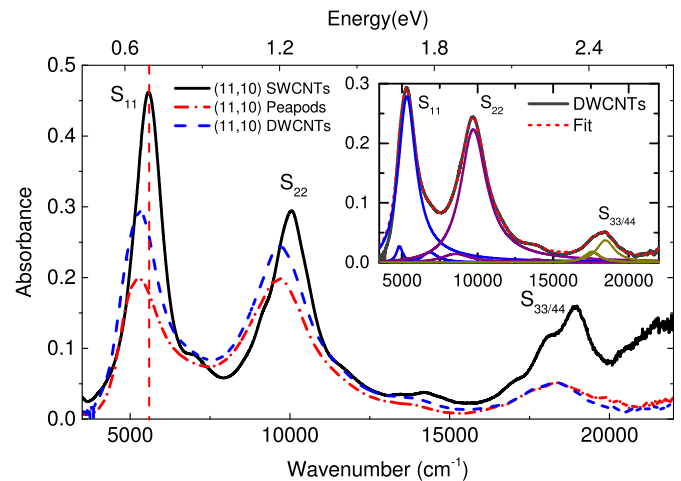


Fig. 5. Background-subtracted absorbance spectra of (11,10) SWCNTs, (11,10) peapods, and (11,10) DWCNTs together with the fit of the absorption bands using Lorentzian oscillators. The absorption bands S_{ij} corresponds to the i th optical transition in the semiconducting SWCNTs. The vertical, dashed line marks the position of the S_{11} band in (11,10) SWCNTs. Inset: Absorbance spectrum for the (11,10) DWCNTs film together with the fit of the absorption bands using Lorentzian oscillators. (A colour version of this figure can be viewed online.)

DWCNTs derived from C_{60} peapods are associated with the RBMs of the inner tubes [27,64]. According to the Kataura plot [35] the 2.4 eV excitation energy matches well with the outer tubes via the energy spacing S_{33} . On the other hand, the 2.4 eV energy is unsuitable for resonant enhancement of inner tubes; only the smaller inner tubes with diameter ~ 0.7 – 0.8 nm are resonant with this excitation energy via the energy spacing S_{22} . Therefore, the observed quite low intensity of the RBM of the inner tubes is expected with the 2.4 eV laser excitation [65,31]. The inner diameter d of the (11,10) DWCNTs was calculated according to the above relation and amounts to $d \approx 0.68$ nm. The obtained value for d is very close to the diameter of the C_{60} fullerene molecules.

The background-subtracted absorbance spectrum of the (11,10) DWCNTs free-standing film is plotted in Fig. 5 in comparison with

(11,10) SWCNTs and (11,10) peapods. The absorption bands were fitted with Lorentzian oscillators (see the inset of Fig. 5) and the contributions are listed in Table 1. One finds that the absorption bands of the (11,10) DWCNTs are red shifted (≈ 30 meV for S_{11} , 40 meV for S_{22} , and 57–60 meV for $S_{33/44}$) compared to the (11,10) SWCNTs, but blue shifted (by ≈ 30 meV for S_{11} , 10 meV for S_{22} , and 10–13 meV for $S_{33/44}$) compared to the (11,10) peapods. The blue shift of the absorption bands in the case of (11,10) DWCNTs compared to (11,10) peapods could be attributed to the decrease of the internal dielectric constant. The dielectric constant of SWCNTs is ~ 3.2 [66] which is less than the dielectric constant of the C_{60} molecules. Furthermore, from Fig. 5 one can see that the spectral weight of the (11,10) DWCNTs absorption bands is increased compared to the (11,10) peapods, but it is still less than that found for the empty (11,10) SWCNTs. Theoretical [67] and experimental [5,68] studies on the charge transfer between the outer and inner tubes in DWCNTs demonstrated that electrons are transferred from the outer wall to the inner one. Therefore, the decrease of the spectral weight of the S_{11} and S_{22} compared to the empty tubes is related to the charge transfer from the outer to the inner tube, and hence electrons are removed from the van Hove singularities in the occupied density of states in the outer tube.

From the above discussion we can conclude that the (11,10) SWCNTs provide a clean environment for studying the interaction between the C_{60} molecules and the nanotubes. In contrast, the absorption bands in multi-chirality SWCNTs are much more complicated, and a precise identification of their modifications with filling is difficult.

4. Conclusion

In conclusion, the interaction between C_{60} fullerene molecules and (11,10) SWCNTs has been characterized in terms of the optical transitions. For (11,10) SWCNTs, the observation of three strong S_{11} , S_{22} , and $S_{33/44}$ absorption bands with the absence of M_{11} metallic contributions reflects the clean and well-defined environment of the (11,10) SWCNTs for studying the interaction between the C_{60} molecules and the nanotubes. The absorption bands of the multi-chirality SWCNTs are red-shifted compared to the (11,10) SWCNTs. This red shift could be attributed to the bundling effect in the multi-chirality SWCNTs. Compared to the empty (11,10) SWCNTs, the optical absorption bands for the (11,10) peapods show a redshift (≈ 60 meV for S_{11} , 50 meV for S_{22} , and 67–70 meV for $S_{33/44}$) and spectral weight reduction in the case of the S_{11} and S_{22} absorption bands. This redshift is explained in terms of an increase of the internal dielectric constant in peapods compared to empty nanotubes. The reduction of the spectral weight of the S_{11} absorption band is attributed to the electron transfer from the (11,10) SWCNTs to the encapsulated C_{60} molecules. This idea is supported by the transformation of the (11,10) peapods to DWCNTs by the laser irradiation method. The formation of the DWCNTs was confirmed by Raman spectroscopy. The inner tubes' diameters in the DWCNTs amounts to ≈ 0.68 nm, as derived from the RBM frequencies. The transformation of the fullerene molecules to an inner tube partially reduces the internal dielectric constant and the charge transfer.

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