Crystal structure of LaTiO_{3.41} under pressure

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The crystal structure of the layered, perovskite-related LaTiO_{3,41} (La₅Ti₅O_{17+ δ}) has been studied by synchrotron powder x-ray diffraction under hydrostatic pressure up to 27 GPa (*T*=295 K). The ambient-pressure phase was found to remain stable up to 18 GPa. A sluggish, but reversible phase transition occurs in the range 18–24 GPa. The structural changes of the low-pressure phase are characterized by a pronounced anisotropy in the axis compressibilities, which are at a ratio of approximately 1:2:3 for the *a*, *b*, and *c* axes. The possible effects of pressure on the electronic properties of LaTiO_{3,41} are discussed.

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I. INTRODUCTION

LaTiO_{3.41} belongs to a series of perovskite-related layered compounds of the composition $A_nB_nO_{3n+2}$.¹⁻⁴ LaTiO_{3.41} is a n=5 member of this class, and thus it may alternatively be denoted as La₅Ti₅O_{17+ δ}. Its monoclinic crystal structure is illustrated in Fig. 1. The samples studied here and in previous works^{1,2,5} have a slight oxygen excess of 0.3% and are therefore denoted as LaTiO_{3.41}, instead of LaTiO_{3.40}.

As a consequence of the oxygen-rich composition of LaTiO_{3.41} (compared to LaTiO₃), its Ti-3*d*-derived electronic bands are only partially occupied (Ti $3d^{0.18}$). Electrical transport measurements^{2,5} on LaTiO_{3,41} revealed strongly anisotropic properties and a quasi-1D metallic behavior. The dc resistivity along the b direction and perpendicular to the ab planes showed semiconducting behavior in the temperature range of 4-290 K. Along the *a* direction a metal-like temperature dependence of the conductivity was reported for the temperature range of 60–200 K. Above 200 K, a slight decrease in resistivity with increasing temperature was observed, indicating conduction due to polaron hopping.^{2,5} Below 60 K the resistivity increased steeply, corresponding to an electronic activation energy of ~ 8 meV. The apparent opening of an electronic band gap has been discussed in terms of a possible Peierls distortion of this quasi-1D system.2

A study of the mid- and far-infrared optical properties of $LaTiO_{3.41}$ has recently corroborated its quasi-1D metallic character.⁵ Furthermore, signatures of polaronic charge carriers were found in the form of a mid-infrared band with pronounced temperature-dependent changes. The polaron scenario is in overall agreement with the experimental results obtained so far. The exact nature of the polaronic carriers, however, still needs to be clarified.

The application of high hydrostatic pressure provides a tool to tune the structural and electronic properties of LaTiO_{3.41}. On the one hand, this may offer a possibility to drive the system more into a quasi-1D metallic regime, i.e., over a wider range of temperatures. On the other hand, and more importantly, it allows for investigation of the nature of the charge carriers, in particular their polaronic character, as was detailed by Goddat *et al.*⁶

Here we explore the structural changes of $LaTiO_{3.41}$ under hydrostatic pressures up to 27 GPa. The crucial question of the structural stability of the ambient-pressure phase is addressed. The pressure-induced structural changes of the lowpressure phase are analyzed and the anticipated changes in the electronic system discussed. We aim at providing a basis for the interpretation of future investigations of the physical properties of LaTiO_{3.41} at high pressures.

II. EXPERIMENTAL DETAILS

The structural properties of LaTiO_{3.41} under pressure were studied up to 27 GPa and ambient temperature by monochromatic (λ =0.4176 Å) x-ray powder diffraction at the European Synchrotron Radiation Facility (ESRF, Grenoble, beamline ID9A). A crystal was ground finely and some powder placed into a diamond anvil cell (DAC) for pressure generation. Nitrogen served as a pressure-transmitting medium to provide nearly hydrostatic conditions. Diffraction patterns were recorded with an image plate detector and then integrated⁷ to yield intensity vs 2θ diagrams. The DAC was rotated by $\pm 3^{\circ}$ during the exposure to improve the powder averaging. Pressures were determined with the ruby luminescence method.⁸ The diffraction diagrams were analyzed with the Rietveld method, i.e., whole pattern fitting, using the GSAS software.^{9,10} The synthesis of the LaTiO_{3,41} material has been described elsewhere.²



FIG. 1. The monoclinic crystal structure (Ref. 1) of LaTiO_{3.41} at ambient conditions (space group $P2_1/c$, Z=4) can be represented by slabs of corner-sharing TiO₆ octahedra. Within the slabs of thickness d_1 , the octahedra are connected and tilted cooperatively in the same fashion as in the more familiar GeFeO₃-type distorted perovskites such as LaTiO₃. La ions are located in the voids between the octahedra.

III. RESULTS AND DISCUSSION

Figure 2(a) shows x-ray diffraction diagrams of LaTiO_{3.41} for increasing pressures up to 24 GPa. At pressures above 2 GPa additional reflections (mostly weak), due to various phases of solid nitrogen, were observed. Up to 18 GPa, there are no discontinuous changes in the diffraction patterns of LaTiO_{3.41}.

At pressures above 18 GPa, a structural phase transition is evidenced by the appearance of additional reflections. The transition is completed at 24 GPa. There is a continuous evolution of the low-angle (002) reflection of the low-pressure phase towards a corresponding peak in the high-pressure diagrams [inset of Fig. 2(a)]. It suggests that the long *c* axis (c_0 =31.5 Å) is preserved across the phase transition. As a result, there are a large number of overlapping reflections, so that a unique determination of the high-pressure unit cell has not been possible. The structural phase transition is reversible.

A rather small increase in the widths of the reflections $(\sim 6\%)$ before and after the $0 \rightarrow 27 \rightarrow 0$ GPa pressure cycle indicates the creation of only a small amount of defects during the phase transformations. It suggests that the phase transition may represent a distortion of the low-pressure crystal structure, rather than a reconstructive transition.

The lattice parameters of the low-pressure phase as a function of pressure were determined from Rietveld-type fits of the diffraction diagrams. This approach was chosen over a direct determination on the basis of the peak positions because of the massive overlap of reflections [Fig. 2(b)]. The lattice parameters were determined up to 18 GPa as shown in Fig. 3(a). The compression under hydrostatic pressure is



FIG. 2. (a) Diffraction diagrams of LaTiO_{3.41} at high pressures (λ =0.4176 Å). The diffractogram of the high-pressure phase at 24 GPa shows a line at 1.65° (inset), corresponding to the (002) reflection of the ambient-pressure phase. (b) X-ray powder diffraction diagram ($I_{\rm obs}$) of LaTiO_{3.41} at 10.0 GPa and difference curve ($I_{\rm obs}-I_{\rm calc}$) for a partial Rietveld refinement (La and Ti positions only). Markers show the calculated peak positions.

rather anisotropic. The relative compressibilities of the a, b, and c directions are at a ratio of approximately 1:2:3.

The inset of Fig. 3(b) shows that the monoclinic angle increases slightly (from 97.17 to 97.43) with increasing pressure up to 18 GPa. It can be determined in two ways: either directly from the refinements (i.e., from the peak positions), or from the ratio of the lattice parameters *a* and *c*. The latter is possible because there are two identical building blocks per unit cell that are shifted by a/4 along the *a* direction, with respect to each other [Fig. 1]. As a consequence, the monoclinic angle β is fully determined by the lattice parameters *a* and *c* through β =arccos(a/2c). The values of β calculated in this way have much smaller uncertainties than those derived directly from the Rietveld refinements [see Fig. 3(b)].

From the lattice parameters we calculate the unit cell volume as a function of pressure as shown in Fig. 3(b). The experimental data are well represented by the Murnaghan relation¹¹ $V(P) = V_0[(B'/B_0)P+1]^{-1/B'}$ with the bulk modulus $B_0 = 142.2(11)$ GPa and its derivative B' = 4.3(2) at zero pressure. Here the ambient-pressure unit cell volume was kept fixed at the experimental value of $V_0 = 1361.5(2)$ Å.³ The ambient-pressure unit cell volume determined here is ~0.4% larger than reported previously.¹



FIG. 3. Structural parameters of LaTiO_{3,41} as a function of pressure. (a) Lattice parameters a, b, c, as well as estimates of the slab thickness d_1 and interlayer separation d_2 , normalized to their respective zero-pressure values. The zero-pressure values of d_1 and d_2 amount to $d_{10} \approx 11.14$ Å and $d_{20} \approx 4.50$ Å, respectively. See text for details. (b) The experimental pressure-volume data can be represented by a Murnaghan equation of state (solid line). The inset shows the variation of the monoclinic angle with pressure. Solid symbols represent angles determined from the lattice parameters a and c; open symbols refer to the results of the refinements.

In order to explore the origin of the anisotropic compressibility, it is worthwhile to estimate, as a function of pressure, the thicknesses d_1 and d_2 of the LaTiO₃-type slabs and the separating layers, respectively [Fig. 1]. A measure of these quantities is provided by the *z* coordinates of the Ti(5) and Ti(6) ions that are adjacent to the oxygen-rich layers. Therefore, an attempt was made to determine the atomic positions of the heavier atoms La and Ti by means of Rietveld refinements. Here, one should bear in mind that the unit cell of LaTiO_{3.41} contains 20 atoms each of La and Ti with 15 La atomic coordinates and 12 Ti coordinates to be determined. This is at the borderline of what can be extracted from highpressure x-ray powder diffraction data.

Figure 4 displays the variation with pressure of the Ti(5)and Ti(6) *z* coordinates. It is evident that the scatter in the values determined directly from the refinements (smaller open symbols) originates primarily from a correlation of



FIG. 4. Evolution of the Ti(5) and Ti(6) z coordinates as a function of pressure. Open squares and diamonds represent the Ti(5) and Ti(6) data, respectively, as determined in the Rietveld refinements. These data scatter in a correlated fashion. The circles show the averaged z coordinates of Ti(5) and Ti(6). The solid line represents a linear fit to the data $[z(P) \approx 0.1781 + 0.00029P/GPa]$, with the outlier at 14 GPa (open circle) not being taken into account. The error bars mark three times the uncertainties reported by the refinement program GSAS.

larger Ti(5) with smaller Ti(6) z positions. At ambient conditions, the Ti(5) and Ti(6) ions are located at $z = 0.1781 \pm 0.0006$, i.e., almost within a common plane. This is very similar to the LaTiO₃ case, where the corresponding Ti atoms are placed strictly in a single plane. Altogether, assuming equal z parameters for the Ti(5) and Ti(6) atoms should provide a good approximation to the actual situation. The averaged Ti(5,6) z coordinates show a clear variation with pressure (see Fig. 4).

From the Ti(5,6) *z* coordinates we can estimate the thicknesses, d_1 of the LaTiO₃-type slabs, and d_2 of the separating layers (cf. Fig. 1). Figure 3(a) shows that the variation of d_1 with pressure is comparable to that of the crystallographic *a* and *b* directions, while the interlayer spacing d_2 is much more compressible. The large compressibility along *c* therefore results from the highly compressible oxygen-rich layers, where the rather strong Ti—O bonds are missing. This large difference in compressibilities also explains the increase of the monoclinic angle with pressure [inset of Fig. 3(b)].

As a check on the plausibility of the estimated d_1, d_2 values we have calculated the bulk modulus of the LaTiO₃-type layers; it amounts to $B_0 \approx 190$ GPa. While there seems to be no experimental data on the bulk modulus of LaTiO₃ available for comparison, our estimate is close to the bulk modulus $B_0=194(3)$ GPa of perovskite-type YTiO₃.¹² Hence, the Ti(5,6) positions and the layer thicknesses derived therefrom can be considered sufficiently accurate to gain some insight into the structural changes within the unit cell.

Additional information on the structural changes near the layer boundaries can be obtained from the La(4) and La(5) positions. At ambient conditions both of these ions are



FIG. 5. Evolution of the La(4) and La(5) atomic coordinates y and z as a function of pressure. Closed (open) symbols refer to refinement results, where preferred orientation effects were (were not) taken into account. The error bars mark three times the uncertainties of the La-y coordinates reported by the refinement program GSAS. For clarity, they are shown only for one data set. In the case of the La-z coordinates, the uncertainties are similar to the symbol size. Lines are guides to the eye.

shifted from their regular positions (z=1/3) between the TiO₆ octahedra into the interlayer region [Fig. 1]. This displacement is more pronounced for La(5) than for La(4). Figure 5 illustrates that La(4) hardly moves along *c* under pressure, while La(5) shifts towards the LaTiO₃-slab region. With regard to the displacement off the *ac* plane (y=0), La(5) moves continuously towards this plane with increasing pressure while La(4) appears to be slightly more displaced at intermediate pressures. Altogether, pressure forces the La(4) and La(5) ions closer to the "ideal" positions in the *ac* planes, they adopt similar positions along the *c* direction, but they remain displaced into the interlayer region.

With respect to the electronic properties of $LaTiO_{3,41}$ (and any other distorted perovskite structure), the tilting of the TiO₆ octahedra represents key information. A measurement of these tiltings in the case of LaTiO_{3.41} would require an accurate determination of the 51 oxygen atomic coordinates, which does not appear feasible on the basis of an x-ray powder diffraction experiment. It is possible, however, to arrive at a semiquantitative estimate of the tilt angles under pressure on the basis of the lattice parameters and the slab thickness d_1 . The relation between octahedral tilting and orthorhombic distortion of the unit cell is well known for the three-dimensionally-linked perovskites of the ABO_3 type.^{13–15} The same scheme can be applied to the LaTiO₃-type slabs of LaTiO_{3.41}, where the octahedra are connected and tilted in the same fashion. In the following, $d_1/2$ takes the role of the third lattice parameter besides a and b. On the basis of the equations given by O'Keeffe and Hyde,¹³ we use the relations



FIG. 6. Octahedral tilt angles φ vs pressure derived from the lattice parameters *a* and *b* and the slab thickness d_1 , according to Eqs. (1) and (2).

$$\varphi(a,b,d_1) = \arccos\left(\frac{\sqrt{8b^2}}{ad_1}\right) \tag{1}$$

and

$$\varphi(a,b) = \arccos(6(b/a)^2 - 2) \tag{2}$$

to determine the octahedral tilt angles shown in Fig. 6. The geometrical constraints of the cooperative octahedral tilting that are at the origin of these relations and the definition of φ are detailed in Ref. 13. Evaluation of the tilt angle on the basis of Eq. (1) with three lattice parameters is generally less sensitive to distortions of the octahedra than the twoparameter determination. The important result is that both approaches indicate a significant increase in the average octahedral tilt angle φ with increasing pressure. Roughly, the tilting angle doubles at 18 GPa compared to ambient conditions.

The indication of a pressure-induced increase in octahedral tilt represents an interesting difference in comparison to other transition-metal perovskites, such as $LaMnO_3$,^{16,17} PrNiO₃,¹⁵ and NdNiO₃,¹⁸ which exhibit a reduction in the tilt. While there are certainly no universal pressure-induced changes in the GdFeO₃-type perovskite compounds, as has been pointed out before,¹⁵ there may exist a systematic behavior across the series of *RTO*₃ rare-earth transition-metal oxides that appears worthwhile to be explored.

Stability of LaTiO_{3.41} under pressures of up to 18 GPa opens way to tune its electrical transport properties. On the one hand, one can vary the itinerancy of the system that is determined by the Ti—O bond lengths and Ti—O—Ti bond angles. A reduction in bond lengths results in a larger bandwidth W of the electronic bands derived from the t_{2g} orbitals and an enhanced itinerancy of the system, while a decrease in the bond angles (increased octahedral tilt) has the opposite effect. Hence, we have here two competing

pressure-induced changes that will affect the itinerancy of the system. These changes are most important for the quasi-1D metallic state that exists in the temperature range of 60-200 K at ambient pressure. On the other hand, polaron binding energies are generally expected to decrease with increasing pressure due to the stiffening of the lattice.⁶ Therefore, high-pressure optical and/or electrical transport experiments are promising approaches to test the hypothesis on polaronic conductivity in the quasi-1D LaTiO_{3 41}.

In conclusion, we have studied the crystal structure of $LaTiO_{3,41}$ by synchrotron x-ray powder diffraction up to 27 GPa. The ambient-pressure phase remains stable up to

18 GPa (at 295 K). Above 18 GPa, a sluggish phase transition occurs, which is completed at 24 GPa. The low-pressure phase is characterized by a pronounced anisotropy of the axis compressibilities at a ratio of approximately 1:2:3 for the *a*, *b*, and *c* axes. The anisotropy can be related to rather compressible oxygen-rich layers that separate the perovskite-type slabs and to variations in the tilt angles of the TiO₆ octahedra. Stability of LaTiO_{3.41} up to 18 GPa opens way to tune its electronic transport properties over a relatively large range. It motivates further electrical transport and optical experiments to study the unusual transport properties of LaTiO_{3.41} in more detail.

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at least two ways, i.e., without and with preferred orientation corrections applied. The differences in the results mark the experimental uncertainty. Preferred orientation effects were modeled using spherical harmonics. A simple pseudo-Voigt line shape was sufficient to describe the reflection profiles up to the highest pressures.

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