### PHASE AND GLASS TRANSITIONS IN $(KI)_{1-x}(NH_4I)_x$

I.FEHST, S.L.HUTTON, R.BÖHMER, A.LOIDL and S.HAUSSÜHL<sup>1</sup> Institut für Physik, Johannes Gutenberg-Universität Mainz, 6500 Mainz, Federal Republic of Germany <sup>1</sup>Institut für Kristallographie, Universität zu Köln, 5000 Köln, Federal Republic of Germany

<u>Abstract</u>: Phase transition temperatures and freezing temperatures have been determined in  $(KI)_{1-x}(NH_4I)_x$  for concentrations  $0 \le x_{sol} \le 1$  using dielectric spectroscopy. From these results a schematic x,T-phase diagram has been deduced. In addition some preliminary results on deuterated samples are presented.

#### **INTRODUCTION**

It is known since a long time that KI and NH<sub>4</sub>I show complete solid state miscibility.<sup>1</sup> Only recently however the potential of this compound to form an orientational glass state <sup>2</sup> has been recognized by Bostoen et al.<sup>3</sup> Based on incoherent inelastic neutron scattering experiments these authors argued that the low temperature excitation spectra of  $(KI)_{1-x}(NH_4I)_x$  mixed crystals are governed by long ranged interaction forces, i. e. not only by the octupoleoctupole exchange between the tetrahedrally shaped ammonium ions. Indeed, subsequent dielectric<sup>4</sup>, hypersonic<sup>5</sup> and Raman scattering<sup>6</sup> experiments confirmed this conjecture. In particular the elastic shear constant  $c_{44}$  showed a softening at low temperatures and the dielectric constant of these crystals exhibited a Curie-Weiss behavior. The latter indicated the existence of a large dipole moment ( $\mu = 1.4D$ ) of the NH<sup>+</sup><sub>4</sub> ion when embedded in a lattice with rocksalt symmetry. No dipole moment is detectable in the low temperature CsCl-type phase of  $NH_4I$  and is presumably also absent in samples where the ammonium concentration is still high enough to drive the martensitic NaCl  $\rightarrow$ CsCl transformation. However the phase diagram of this binary system is not

known. Therefore we have investigated five mixed crystals covering the entire concentration range 0 < x < 1 using dielectric methods. This allowed the determination of important features of the phase diagram of  $(KI)_{1-x}(NH_4I)_x$  as well as of the critical concentration  $x_c$  up to which the mixed crystals maintain their rocksalt structure down to lowest temperatures. In this concentration regime the strong and anisotropic dipolar interactions together with the random distribution of the  $NH_4^+$  dipoles on the lattice sites favor the establishment of a dipolar glass state<sup>4</sup>.

We will briefly discuss the effect of deuteration on the freezing process, which is of interest also in other dipolar systems like the RADP-type proton glasses<sup>2</sup> or in the octupolarly disordered methane-krypton solid solutions<sup>7</sup>.

### **EXPERIMENTAL**

Small crystals of  $(\text{KI})_{1-x}(\text{NH}_4\text{I})_x$  with  $x_{sol} = 0.7$ , 0.8, 0.9 were grown from aqueous solution. Here  $x_{sol}$  characterizes the NH<sub>4</sub>I concentration in the solution. In order to perform dielectric measurements on these relatively small ingots they were powdered and filled into a capacitor for which the geometrical capacity was 5 pF. The filling factor of the configuration was estimated to be about 25 %. Measurements on gold plated single crystals were performed for concentrations  $x_{sol} = 0.5$  (actual ammonium fraction x = 0.43) and 1<sup>4</sup>, as well as for a deuterated sample  $(\text{KI})_{1-x}(\text{ND}_4\text{I})_x$  with  $x_{sol} = 0.5$ . The complex dielectric constant  $\epsilon = \epsilon_1 - i\epsilon_2$  was measured in the frequency range  $10^{-2}$  Hz to  $10^7$  Hz using two and four wire impedance techniques as described previously<sup>8</sup>.

### RESULTS AND DISCUSSION

In pure NH<sub>4</sub>I the real part of the dielectric constant  $\epsilon_1$  increases with decreasing temperature until at 242 K the martensitic phase transition causes a sudden drop in  $\epsilon_1$ <sup>4</sup>. Lowering the concentrations to  $x_{sol} = 0.9$  respectively 0.8, the general behavior is the same as in the pure compound (see fig.1): the drop in  $\epsilon_1$ 

smeares out and is shifted to lower temperatures. The step in  $\epsilon_1$  broadens with decreasing concentration of the ammonium ions. This indicates that the increasing distortion of the lattice by replacing K<sup>+</sup> for NH<sup>+</sup><sub>4</sub> ions drives the lowering of the phase transition temperature. The lack of relaxation in the dielectric loss  $\epsilon_2$  of the samples with  $x_{sol} > 0.8$  indicates that the step in  $\epsilon_1$  is induced by a phase transition. In the sample  $x_{sol} = 0.8$  the imaginary part  $\epsilon_2$  shows a slight and also smeared out maximum without dispersion, which indicates the onset of a glass like behavior as observed more distinctly for the samples with lower concentrations<sup>4</sup>. In addition in temperature cycling experiments a hysteresis of about 60 K was observed which is quite large and may be compared to that observed for example in NaCN by X-ray diffraction<sup>9</sup>. This hysteresis can be interpreted in terms of the still existing martensitic phase transition. These observations suggest that  $x_{sol} = 0.8$  is near the critical concentration of this binary system.



FIGURE 1 Schematic phase diagram of the binary system  $(KI)_{1-x}(NH_4I)_x$ .  $\epsilon_1$  versus temperature is given as an insert for several concentrations  $x_{sol}$ . The transition temperatures for  $x_{sol} = 0.9$  and 0.8 are determined by the corresponding cusps in  $\epsilon_1$ . The real parts of the dielectric constants  $\epsilon_1$  for samples with the concentrations  $x_{sol} = 0.7$  and 0.5 (the actual concentration in this crystal is x = 0.43) are given in the insert of fig. 1, respectively in fig. 2. For these crystals a Curie-Weiss behavior  $\chi = C/(T - \Theta)$  of the static susceptibility  $\chi = (\epsilon_1 - \epsilon_{\infty})/4\pi$  was observed at high temperature. By taking into account the lattice polarization contributions  $\epsilon_{\infty} = 4.68 + T/700$  of pure KI<sup>10</sup> we calculated the Curie constant C = 117 K and the Weiss temperature  $\Theta = -163$  K. The negative value of  $\Theta$  is a measure of the antiferroelectric dipolar correlations.

At low temperatures the real part of the dielectric constant  $\epsilon_1$  drops to lower values and is indicative for glassy freezing as studied previously in more detail for a  $(KI)_{1-x}(NH_4I)_x$  sample with  $x_{sol} = 0.5$ . The imaginary part  $\epsilon_2$  shows maxima. From the temperature shift of the loss maxima the hindering barrier E and the attempt frequency  $\nu_0$  were estimated by parameterizing the experimental results with an Arrhenius law.



FIGURE 2 Dielectric constant  $\epsilon_1$  versus temperature T for  $(KI)_{1-x}(NH_4I)_x$ . The concentrations are  $x_{sol} = 0.2$  and 0.5 The corresponding concentrations in the crystals are x = 0.14and 0.43. The measuring frequency was 100 kHz. In order to find out how deuteration effects the freezing we investigated a deuterated sample with  $x_{sol} = 0.5$ . We find that the general behavior of the dielectric constant of the deuterated (D) sample was similar to that observed in the protonated (H) sample. In the D-compound also a Curie-Weiss behavior was seen with a Curie constant of C = 125 K and a Weiss temperature of  $\Theta = -131K$ . Within the experimental error this leads to the same dipole moment  $\mu$  as in the protonated sample. A difference was observed in the temperatures where the cusps in  $\epsilon_1$  and the loss maxima appeared. The temperatures are shifted to lower values which is the opposite of what is usually observed upon deuteration<sup>7</sup>. The temperature shift of the loss maxima is different from that observed in the H-sample and gives a significantly lower hindering barrier. A full account of our results will be given elsewhere.

## SUMMARY

In conclusion we have presented preliminary results on protonated and deuterated crystals of  $(KI)_{1-x}(NH_4I)_x$ . For the protonated compounds a schematic x,T-phase diagram is presented. the critical concentration  $x_c$  is close to 0.8. In the rocksalt structure the deuterated compounds reveal a similar magnitude of the dipole moment ( $\mu \approx 1.4$  D). The freezing temperatures in the deuterated samples are shifted to lower values.

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