

## Limited effect of anthropogenic nitrogen oxides on secondary organic aerosol formation

Y. Zheng, N. Unger, A. Hodzic, L. Emmons, Christoph Knote, S. Tilmes, J.-F. Lamarque, P. Yu

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*Supplement of*

## **Limited effect of anthropogenic nitrogen oxides on secondary organic aerosol formation**

**Y. Zheng et al.**

*Correspondence to:* Y. Zheng (yiqi.zheng@yale.edu)

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### **Simulation using “VBS\_agHigh\_isop” scheme.**

The VBS schemes (VBS, VBS\_agHigh, etc.) that we implement into CAM4-chem as described in the main text are based on the published VBS parameterizations from Pye et al. (2010), which does not consider the high-NO<sub>x</sub> pathway for isoprene oxidation. Here we performed an additional simulation “VBS\_agHigh\_isop” to evaluate the effect of this omitted pathway.

In the VBS\_agHigh\_isop scheme, we add the high-NO<sub>x</sub> branch for isoprene-oxidized SOA formation (ISOPO<sub>2</sub>+NO) using the parameters from Lane et al. (2008). The SOA mass yields are 0, 0.00026, 0.0195, 0.013 for C\* of 0.1, 1, 10, 100  $\mu\text{g m}^{-3}$ , respectively, fit using a density of 1.3  $\text{g cm}^{-3}$  to be consistent with Pye et al. (2010). Adding this pathway, the VBS\_agHigh\_isop scheme gives an annual mean SOA burden of  $0.86 \pm 0.05 \text{ Tg[C]}$  and a net SOA production of  $46.8 \pm 1.9 \text{ Tg[C]}$  per year, both of which are about 20% smaller than the VBS\_agHigh scheme, and larger than the two no-aging schemes (2-product and VBS). The VBS\_agHigh\_isop scheme shows no significant advantage over the VBS\_agHigh scheme in model-observation comparisons.

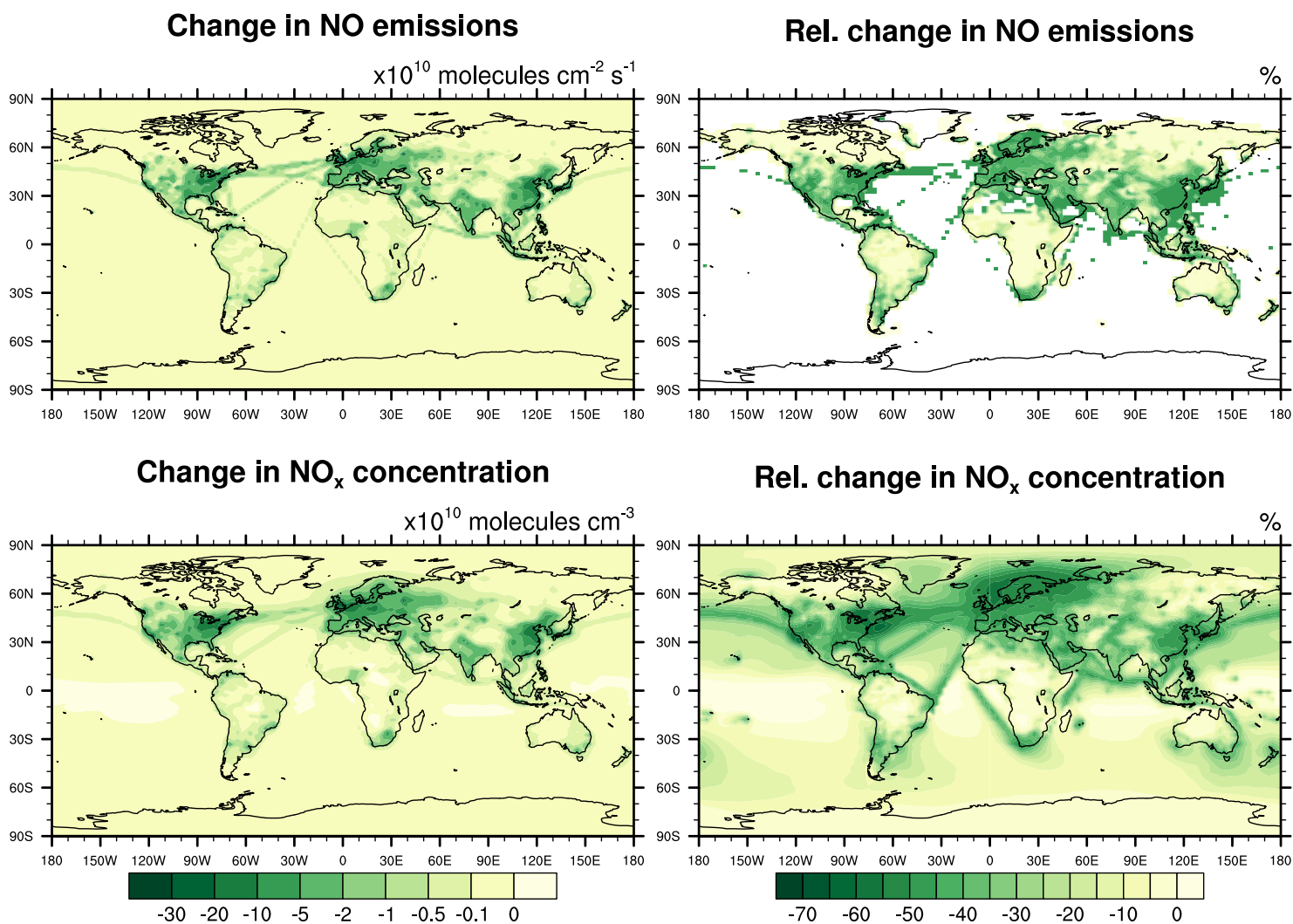
We then performed a sensitivity run using the VBS\_agHigh\_isop scheme, in which anthropogenic NO emissions are reduced by 50%. Results in the southeast US and the Amazon are shown in Fig. S5. SOAM and anthropogenic SOAs through each branch behave similar to the VBS\_agHigh. For isoprene oxidation in the southeast US, the HO<sub>2</sub> pathway increases and NO pathway decrease. While the effect of shifting to high-yield HO<sub>2</sub> pathway is masked due to the reduced OH concentration (about -15%, Fig. S2). The total SOA concentration decreases from  $3.4 \pm 0.5$  to  $3.0 \pm 0.4 \mu\text{g m}^{-3}$ . This  $0.4 \mu\text{g m}^{-3}$ , 10% reduction is smaller than the change in VBS\_agHigh scheme ( $0.5 \mu\text{g m}^{-3}$ , 12%), and is smaller than one standard deviation of multi-year variation, suggesting the SOA response to the NO perturbation is not significant. In the Amazon, all branches show a decrease with reduced NO. The total SOA reduction ( $0.1 \mu\text{g m}^{-3}$ , 2%) is non-significant. These conclusions are consistent with the VBS\_agHigh results.

**Table S1. SOA mass yields using default 2-product scheme (Heald et al., 2008 and references therein).** The reference temperature is 310K for monoterpenes and isoprene, and is 295K for benzene, toluene and xylenes. The enthalpy of evaporation is 42 kJ mol<sup>-1</sup>.

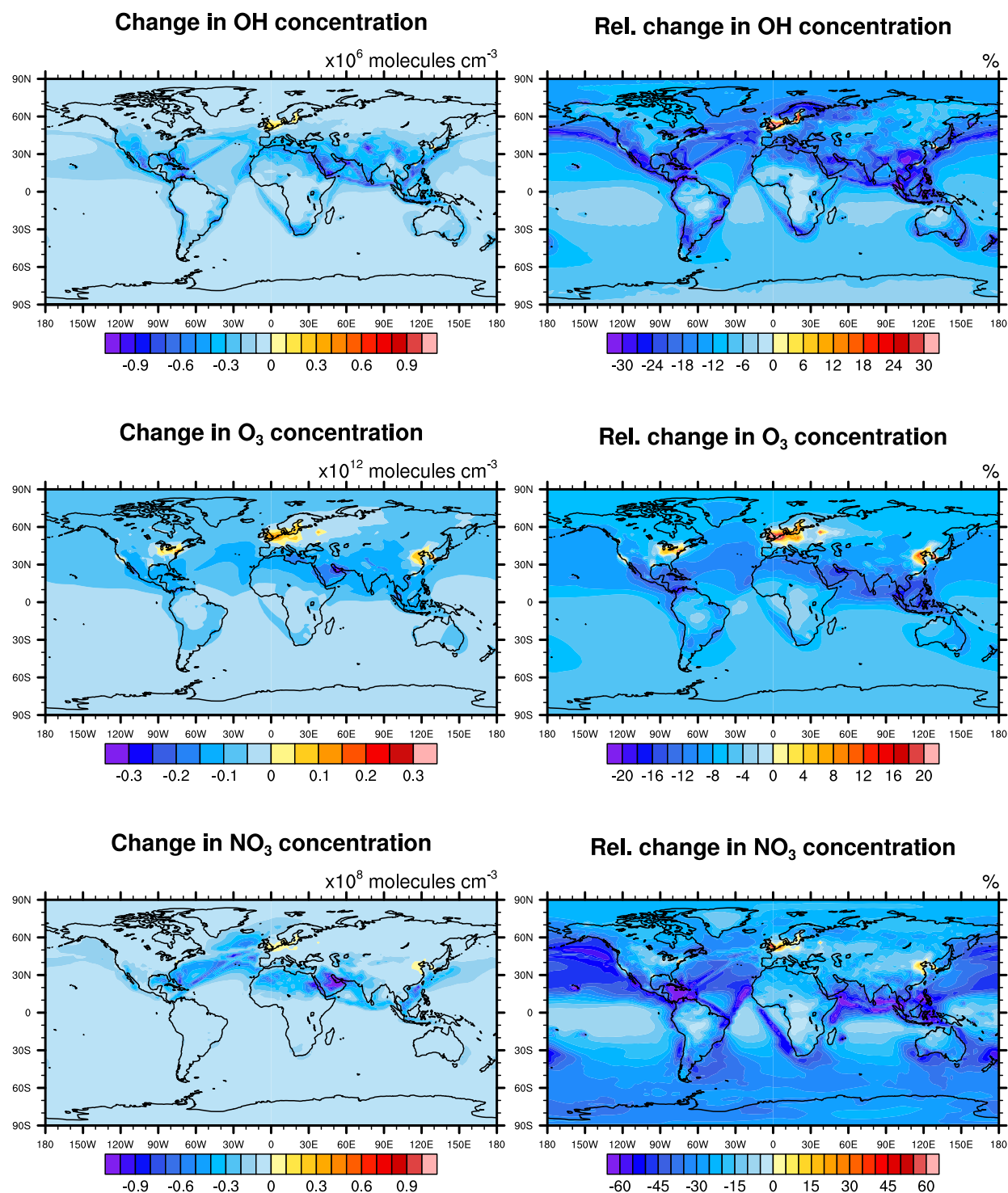
Parent HC	Oxidants	Surrogate product	$\alpha$ and C* (C* in $\mu\text{g m}^{-3}$ )		Yield at 10 (and 1) $\mu\text{g m}^{-3}$
Monoterpenes C <sub>10</sub> H <sub>16</sub>	OH; HO <sub>2</sub>	SOAM C <sub>10</sub> H <sub>16</sub> O <sub>4</sub>	C*=5.435 $\alpha$ =0.067	C*=232.6 $\alpha$ =0.354	0.06 (0.01)
	O <sub>3</sub> ; HO <sub>2</sub>		C*=5.435 $\alpha$ =0.067	C*=232.6 $\alpha$ =0.354	0.06 (0.01)
	NO <sub>3</sub>		C*=61.35 $\alpha$ =1.000	0	0.14 (0.02)
Isoprene C <sub>5</sub> H <sub>8</sub>	OH; HO <sub>2</sub>	SOAI C <sub>5</sub> H <sub>12</sub> O <sub>4</sub>	C*=0.617 $\alpha$ =0.029	C*=116.0 $\alpha$ =0.232	0.05 (0.02)
Benzene C <sub>6</sub> H <sub>6</sub>	OH; HO <sub>2</sub>	SOAB C <sub>6</sub> H <sub>7</sub> O <sub>3</sub>	C*=0.001 $\alpha$ =0.227	0	0.23 (0.23)
	OH; NO		C*=0.302 $\alpha$ =0.044	C*=111.1 $\alpha$ =0.545	0.09 (0.04)
Toluene C <sub>7</sub> H <sub>8</sub>	OH; HO <sub>2</sub>	SOAT C <sub>7</sub> H <sub>9</sub> O <sub>3</sub>	C*=0.001 $\alpha$ =0.235	0	0.23 (0.23)
	OH; NO		C*=2.326 $\alpha$ =0.038	C*=21.28 $\alpha$ =0.074	0.05 (0.01)
Xylenes C <sub>8</sub> H <sub>10</sub>	OH; HO <sub>2</sub>	SOAX C <sub>8</sub> H <sub>11</sub> O <sub>3</sub>	C*=0.001 $\alpha$ =0.205	0	0.20 (0.20)
	OH; NO		C*=1.314 $\alpha$ =0.021	C*=34.48 $\alpha$ =0.061	0.03 (0.01)

**Table S2. SOA mass yields using updated 4-product VBS scheme (Pye et al., 2010 and references therein).** The reference temperature is 295K and the enthalpy of evaporation is 73, 62, 51, 40 kJ mol<sup>-1</sup> for C\* of 0.1, 1, 10, 100 µg m<sup>-3</sup>, respectively.

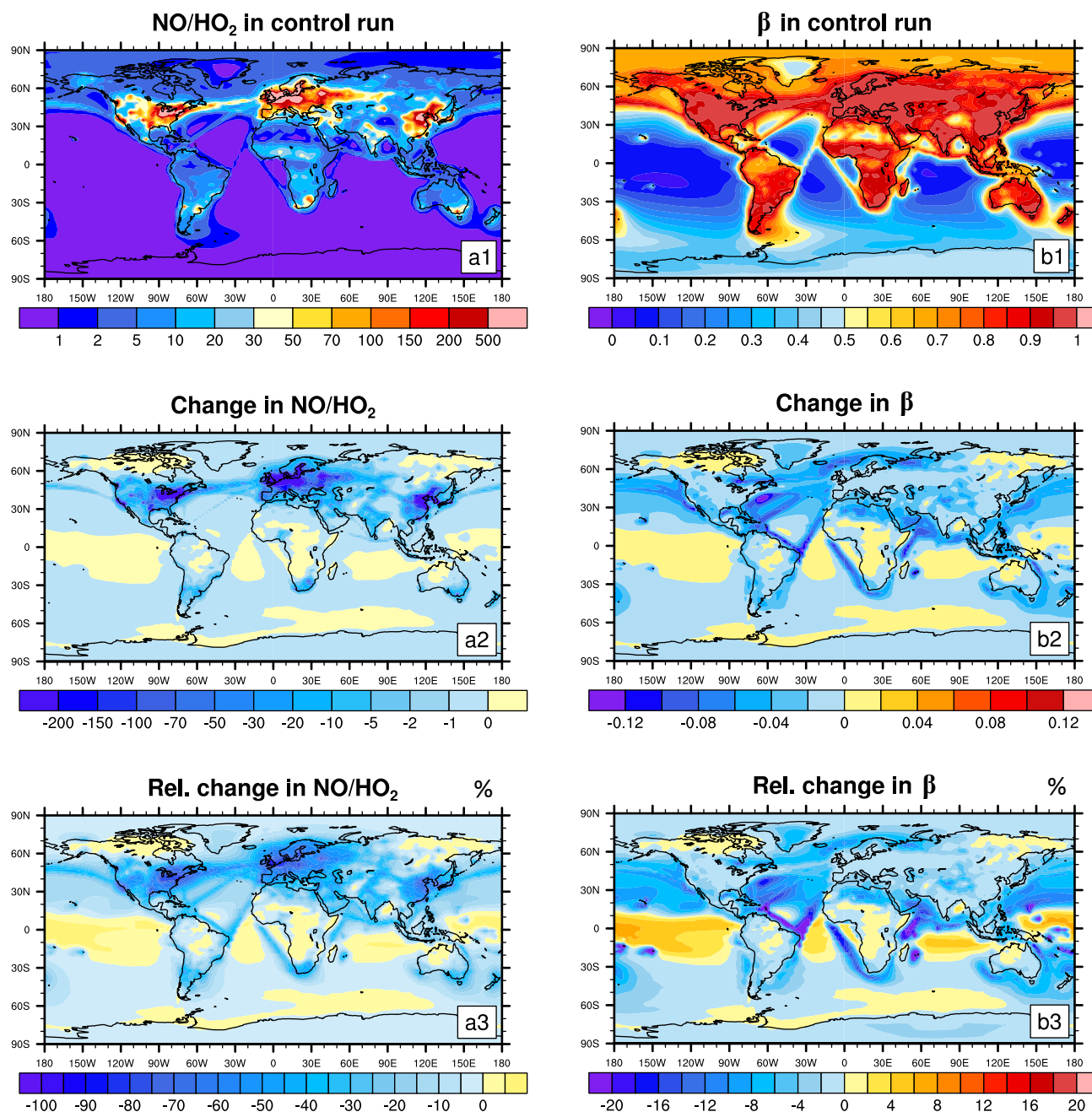
Parent HC	Oxidants	Surrogate product	$\alpha$ for C* (C* in µg m <sup>-3</sup> )				Yield at 10 (and 1) µg m <sup>-3</sup>
			C*=0.1	C*=1	C*=10	C*=100	
Monoterpenes C <sub>10</sub> H <sub>16</sub>	OH; HO <sub>2</sub>	SOAM C <sub>10</sub> H <sub>16</sub> O <sub>4</sub>	0.08	0.019	0.18	0.03	0.19 (0.10)
	OH; NO		0.04	0.0095	0.09	0.015	0.09 (0.05)
	O <sub>3</sub> ; HO <sub>2</sub>		0.08	0.019	0.18	0.03	0.19 (0.10)
	O <sub>3</sub> ; NO		0.04	0.0095	0.09	0.015	0.09 (0.05)
	NO <sub>3</sub>		0	0	0.321	1.083	0.26 (0.04)
Isoprene C <sub>5</sub> H <sub>8</sub>	OH; HO <sub>2</sub>	SOAI	0	0.031	0	0.095	0.04 (0.02)
	NO <sub>3</sub>	C <sub>5</sub> H <sub>12</sub> O <sub>4</sub>	0	0	0.217	0.092	0.12 (0.02)
Benzene C <sub>6</sub> H <sub>6</sub>	OH; HO <sub>2</sub>	SOAB	0.37	0	0	0	0.37 (0.34)
	OH; NO	C <sub>6</sub> H <sub>7</sub> O <sub>3</sub>	0	0.078	0	0.793	0.14 (0.05)
Toluene C <sub>7</sub> H <sub>8</sub>	OH; HO <sub>2</sub>	SOAT	0.36	0	0	0	0.36 (0.33)
	OH; NO	C <sub>7</sub> H <sub>9</sub> O <sub>3</sub>	0	0.032	0.094	0.080	0.08 (0.03)
Xylene C <sub>8</sub> H <sub>10</sub>	OH; HO <sub>2</sub>	SOAX	0.30	0	0	0	0.30 (0.27)
	OH; NO	C <sub>8</sub> H <sub>11</sub> O <sub>3</sub>	0	0.025	0.036	0.090	0.05 (0.02)



**Figure S1. Absolute (left column) and relative (right column) changes of annual mean NO emissions (units:  $\times 10^{10} \text{ molecules cm}^{-2} \text{ s}^{-1}$ ) and simulated surface NO<sub>x</sub> concentration (units:  $\times 10^{10} \text{ molecules cm}^{-3}$ ) in the sensitivity run compared to the control run using VBS<sub>agHigh</sub> scheme. Data are averaged from 2005 to 2009.**

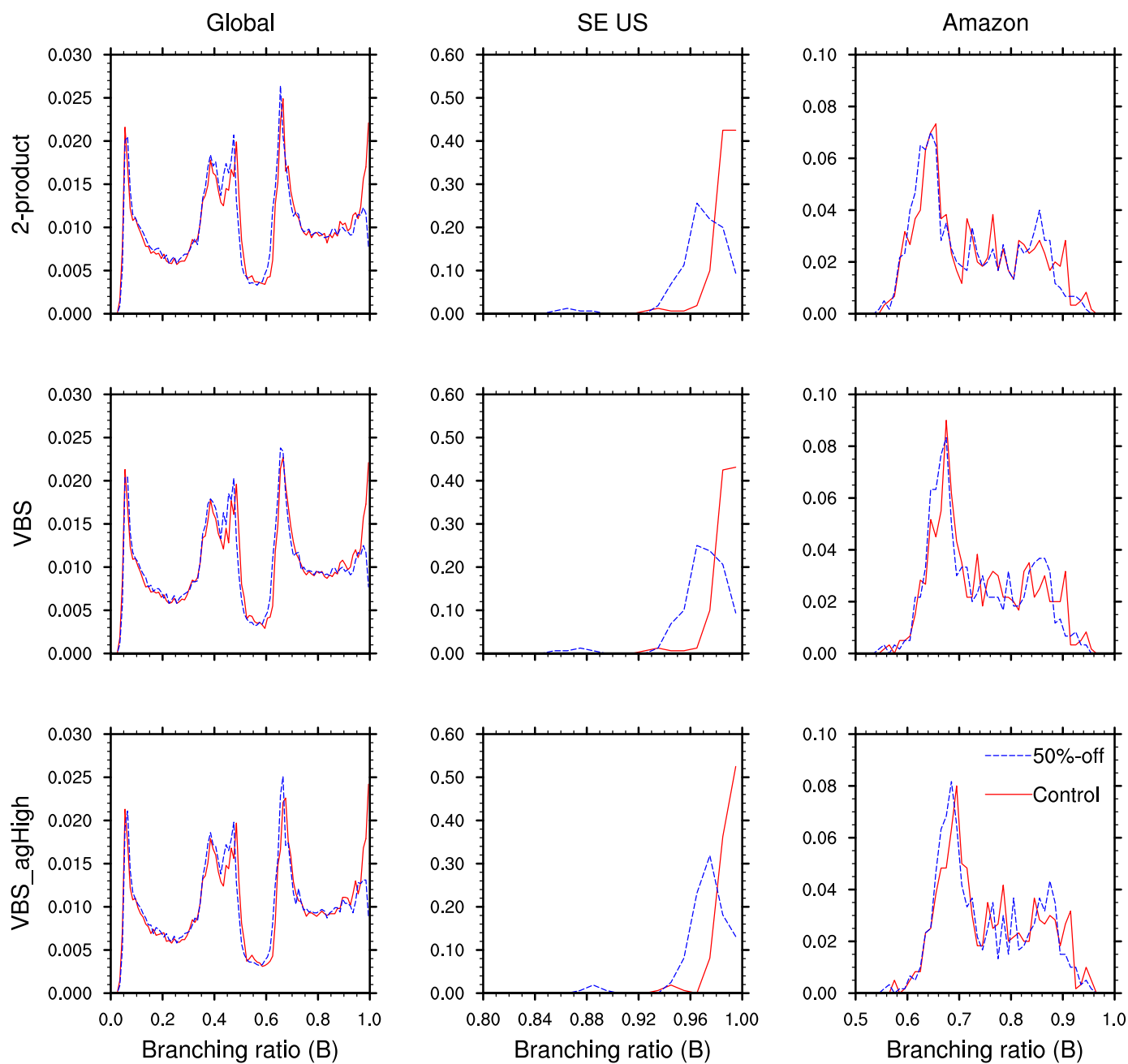


**Figure S2.** Absolute (left column) and relative (right column) changes of simulated annual mean surface OH ( $\times 10^6$  molecules  $\text{cm}^{-3}$ ), O<sub>3</sub> ( $\times 10^{12}$  molecules  $\text{cm}^{-3}$ ) and NO<sub>3</sub> concentrations ( $\times 10^8$  molecules  $\text{cm}^{-3}$ ) in the sensitivity run compared to the control run using VBS\_agHigh scheme. Data are averaged from 2005 to 2009. Note that the range of color bars are different in each subplot.

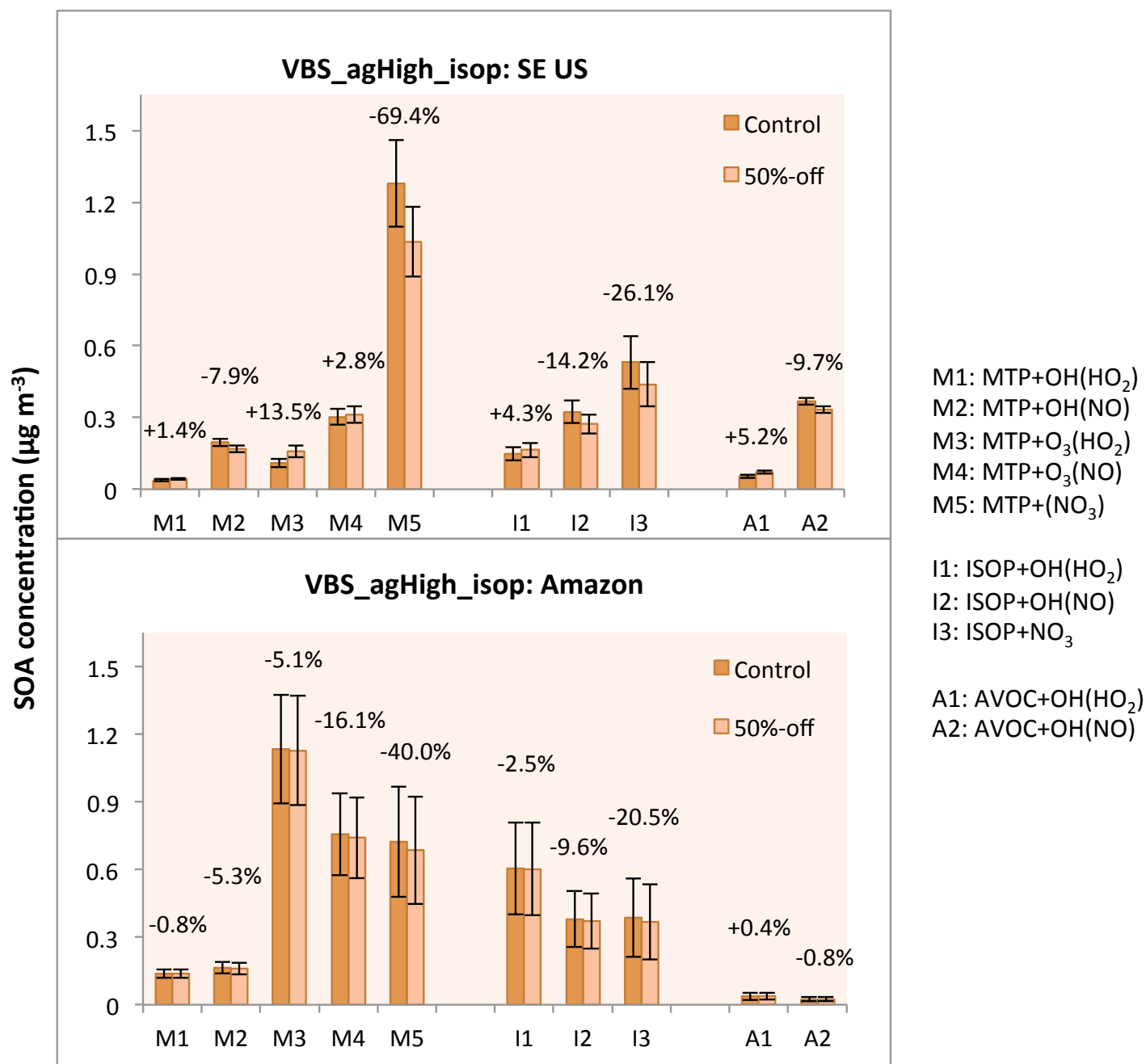


**Figure S3.** Left column: (a1) is the annual mean surface NO-to-HO<sub>2</sub> ratio (NO/HO<sub>2</sub>) in the control run; (a2) and (a3) are the absolute and relative change of NO/HO<sub>2</sub> in the sensitivity run compared to the control run. Right column is similar to the left column but for annual mean surface branching ratio ( $\beta$ ). Simulations use VBS\_agHigh scheme. Data are averaged from 2005 to 2009. Note that the range of color bars are different in each subplot.





**Figure S4. Probability density function of annual mean surface branching ratio ( $\beta$ ) at global scale, in the southeast US and in the Amazon.**



**Figure S5.** Annual mean surface SOA concentration ( $\mu\text{g m}^{-3}$ ) in the control run and the sensitivity run (with 50% anthropogenic NO emission off) from different pathways using the VBS\_agHigh\_isop scheme, averaged over the southeast U.S. [32°-40°N, 95°-77°W] and the Amazon [17°S-5°N, 77°-55°W]. The numbers above each bar denote the relative contributions (%) of each SOA formation pathway to the total SOA concentration change. The sum of all numbers equals -100% due to the overall decrease in total SOA.