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Homogenisation of a chemical degradation mechanism inducing an evolving microstructure

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Abstract

We consider a reaction–diffusion problem in a porous medium, where the reaction causes a local increase or decrease of volume of the solid matrix. For the homogenisation of the resulting system of equations, we employ the method of homogenisation in domains with evolving microstructure. The functions describing the evolution of the microstructure are related to the reaction–diffusion process.

Résumé

Homogénéisation d'un mécanisme du effritement chimique qui cause une évolution de la micro-structure. On considère un problème de la réaction et la diffusion dans un milieu poreux, où la réaction cause une augmentation ou une réduction du volume de la matrix solide. Par l'homogénéisation du système d'équations résultants on emploie la méthode d'homogénéisation dans des domaines avec évolution de la micro-structure. Les fonctions que décrivent l'évolution de la micro-structure ont relié au processus de la réaction et la diffusion.

Key words: Homogenization ; multiscale approach ; reaction–diffusion ; concrete carbonation ; porous media

Mots-clés : Homogénéisation ; approche multiéchelle ; réaction et diffusion ; carbonatation du beton ; milieu poreux

1. Introduction

We consider a reaction–diffusion problem involving two species in a porous medium Ω made up of pore air Ω^a , pore water Ω^w and solid matrix Ω^s . One species, A, diffuses through the pore air and dissolves in

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the pore water. Another species, B, is part of the solid matrix and dissolves into the pore water. The two species may participate in a reaction of the form



where the reaction product C is of low solubility and (after precipitation) can hence be associated with the solid matrix again. The special feature of the problem under consideration is that owing to different molar densities of B and C, the reaction causes a local growing or shrinking of the solid matrix. In turn, the pore air and pore water change. Therefore, the total volume occupied by the porous medium Ω is time-independent while its parts, Ω^a , Ω^w and Ω^s , may evolve owing to the conversion of B to C.

One particular application exhibiting these features is concrete carbonation, where the reactant A is atmospheric carbon dioxide, which reacts with certain concrete constituents, mainly calcium hydroxide (B). Calcium carbonate (C) is produced in this reaction, which causes a decrease of pH in the concrete and an increase of volume of solid matrix since the calcium carbonate takes up more volume than the calcium hydroxide. The drop in pH facilitates the corrosion of reinforcing steel bars and thus decreases the service-life of the structure. More information about concrete carbonation can be found in [1,2,3], e.g., and in [4,5] in the context of multi-scale approaches.

We present a coupled reaction–diffusion system modelling the concentration of A in the two evolving domains occupied by pore air and pore water. Making use of the method of homogenisation in domains with evolving microstructure presented in [6], we use periodic homogenisation techniques, cf. [7,8] e.g., to find the macroscopic system of equations. In order to focus on the evolution induced by the reaction (1) and to keep the problem similar to the abstract problem presented in [6], we assume the reaction rate to be given and we do not explicitly consider the concentration of B or C. The problem under consideration can be viewed as a starting point for more complex reaction scenarios including an unknown for the concentration of B as well as non-linear concentration-dependent reaction rates, such as those considered in [9].

2. The microscopic reaction–diffusion system

In order to upscale the problem using periodic homogenisation techniques, we employ the ideas of homogenisation in domains with evolving microstructure described in [6], i.e. it is assumed that there exists a reference configuration, which is ε -periodic. By this, we mean that there exists a reference cell $Y = (0, 1)^N$ with $\bar{Y} = \bar{Z}^a \cup \bar{Z}^w \cup \bar{Z}^s$ and $Z^a \cap Z^w \cap Z^s = \emptyset$, such that $\Omega_*^i = \bar{\Omega} \cap \text{int} \bigcup_{k \in \mathbb{Z}^N} \varepsilon(\bar{Z}^i + k)$ is a reference configuration of $\Omega^i(t)$, $i \in \{a, w, s\}$.

For simplicity, the geometry at $t = 0$ is assumed to coincide with this reference configuration. Furthermore, the evolution of the three parts of Ω can be described by orientation-preserving mappings $\psi_\varepsilon^i(\cdot, t) : \Omega_\varepsilon^i \rightarrow \Omega^i(t)$, $i \in \{a, w, s\}$, for each $t \in S = (0, T)$. Here, we have written Ω_ε^i for $\Omega^i(0) = \Omega_*^i$ in order to emphasise the ε -periodicity of the geometry. We define $\Psi_\varepsilon^i = \nabla \psi_\varepsilon^i$ and $J_\varepsilon^i = \det \Psi_\varepsilon^i$, $i \in \{a, w, s\}$. For the sake of brevity, we write $\Gamma_\varepsilon = \partial\Omega_\varepsilon^a \cap \partial\Omega_\varepsilon^w$ and $\Gamma_\varepsilon^{ws} = \partial\Omega_\varepsilon^w \cap \partial\Omega_\varepsilon^s$ and we assume $\partial\Omega_\varepsilon^a \cap \partial\Omega_\varepsilon^s = \emptyset$.

In order to keep things simple and to focus on the aspect of the evolution of the microstructure being induced by the reaction, only the concentrations of A in Ω_ε^a and Ω_ε^w are modelled, which we denote by u_ε^a and u_ε^w , respectively. Cf. eq. (2) in [6], the reaction–diffusion system in the reference description is thus given by

$$\partial_t(J_\varepsilon^a u_\varepsilon^a) - \nabla \cdot (J_\varepsilon^a \Psi_\varepsilon^{a-1} D_\varepsilon^a \Psi_\varepsilon^{a-T} \nabla u_\varepsilon^a) = 0, \quad x \in \Omega_\varepsilon^a, t \in S, \quad (2a)$$

$$\partial_t(J_\varepsilon^w u_\varepsilon^w) - \nabla \cdot (\varepsilon^l J_\varepsilon^w \Psi_\varepsilon^{w-1} D_\varepsilon^w \Psi_\varepsilon^{w-T} \nabla u_\varepsilon^w) = -r_\varepsilon f_\varepsilon^r, \quad x \in \Omega_\varepsilon^w, t \in S,$$

$$-(J_\varepsilon^a \Psi_\varepsilon^{a-1} D_\varepsilon^a \Psi_\varepsilon^{a-T} \nabla u_\varepsilon^a) \cdot \nu_\varepsilon^a = (\varepsilon^l J_\varepsilon^w \Psi_\varepsilon^{w-1} D_\varepsilon^w \Psi_\varepsilon^{w-T} \nabla u_\varepsilon^w) \cdot \nu_\varepsilon^w, \quad x \in \Gamma_\varepsilon, t \in S,$$

$$-(J_\varepsilon^a \Psi_\varepsilon^{a-1} D_\varepsilon^a \Psi_\varepsilon^{a-T} \nabla u_\varepsilon^a) \cdot \nu_\varepsilon^a = \varepsilon a_\varepsilon \|\Psi_\varepsilon^{w-T} \nu_\varepsilon^w\| J_\varepsilon^w (u_\varepsilon^a - u_\varepsilon^w), \quad x \in \Gamma_\varepsilon, t \in S, \quad (2b)$$

$$-(\varepsilon^l J_\varepsilon^w \Psi_\varepsilon^{w-1} D_\varepsilon^w \Psi_\varepsilon^{w-T} \nabla u_\varepsilon^w) \cdot \nu_\varepsilon^w = 0, \quad x \in \Gamma_\varepsilon^{ws}, t \in S,$$

together with boundary conditions at the exterior boundary as well as initial conditions. Compared to [6], we have abandoned the hats to denote functions in the reference description and we only have evolution of the domains due to deformation since we do not model concentrations in the solid matrix here.

The scaling factor ε^l associated with D_ε^w , $l \in [0, 2]$, arises from a nondimensionalisation. It is related to the characteristic macroscopic and microscopic lengths as well as the diffusion times. Similar consideration motivate the scaling of the interfacial-exchange coefficient a_ε . More details on the influence and choice of different scaling exponents can be found in [10,9]. We require Ω_ε^a and, if $l = 0$, Ω_ε^w to be connected.

The only functions associated with the evolution of the domains appearing in the system of equations in the reference configuration (2) are Ψ_ε^i and J_ε^i , $i \in \{a, w\}$. We want to relate these to the reaction–diffusion process. The reaction (1) consumes a matrix constituent and produces another one. Therefore, we need to model the evolution of the subdomains depending on the reaction. For this, recall that Ψ_ε^i relates the length and orientation of a material fibre in the reference configuration to its length and orientation in the current configuration and J_ε^i describes the change of volume.

We consider the case where it is reasonable to assume that changes in geometry other than the volumetric changes do not have a large influence on the problem, i.e. we can assume $\Psi_\varepsilon^i \approx \text{Id}$, $i \in \{a, w\}$, in the system of equations (2). For example, this is the case if nearly isotropic diffusion is always to be expected. Since it is reasonable to assume the water incompressible, we have $J_\varepsilon^w \equiv 1$ for all times. Therefore, we are left with deriving an equation for J_ε^a . Because we need to relate the change of pore-air volume to the reaction in the pore water, we only obtain a (non-local) relation of the *averaged quantities*.

We proceed with the derivation of an equation for J_ε^a . The dissolution of B and the precipitation of C are assumed to be instantaneous in order to keep things simple. The reaction rate is given by a function $f_\varepsilon^r(x, t)$, which is bounded independently of ε in $L^2(\Omega_\varepsilon^w \times S)$. We denote by $\tilde{f}_\varepsilon^r(x, y, t)$ the unfolded version of f_ε^r , i.e. $\tilde{f}_\varepsilon^r = \mathcal{T}_\varepsilon(f_\varepsilon^r)$ where \mathcal{T}_ε is the periodic unfolding operator [11], and analogously \tilde{J}_ε^a and \tilde{J}_ε^w . In the cell Y at the point x , the amount (i.e. mass) of constituent B being used up and of constituent C being produced at time t is given by

$$\int_{Z^w} m_B \tilde{J}_\varepsilon^w \tilde{f}_\varepsilon^r \, dy \quad \text{and} \quad \int_{Z^w} m_C \tilde{J}_\varepsilon^w \tilde{f}_\varepsilon^r \, dy, \quad (3)$$

respectively, where m_B and m_C are the molar weights of B and C, respectively. Noting that $\tilde{J}_\varepsilon^w \equiv 1$ and that an increase of total volume of constituents B and C implies a decrease of volume of pore air, we have

$$\frac{d}{dt} \int_{Z^a} \tilde{J}_\varepsilon^a \, dy = -|Z^a| C^m \int_{Z^w} \tilde{f}_\varepsilon^r \, dy \quad \text{where} \quad C^m = \frac{1}{|Z^a|} \left(\frac{m_C}{\rho_C} - \frac{m_B}{\rho_B} \right) \quad (4)$$

for each instant in time, where ρ_B and ρ_C are the densities of constituents B and C, respectively.

Since we do not expect the limit of J_ε^a as $\varepsilon \rightarrow 0$ to vary within one cell (and since we do not have any more information), we *define* $J_\varepsilon^a(x, t)$ to be the constant value determined by (4) in each cell. Hence, an equation for J_ε^a is obtained: find $J_\varepsilon^a: \Omega \times S \rightarrow \mathbb{R}$ such that

$$\partial_t J_\varepsilon^a(x, t) = -C^m \int_{Z^w} \tilde{f}_\varepsilon^r(x, y, t) \, dy, \quad J_\varepsilon^a(0) \equiv 1. \quad (5a)$$

The reaction–diffusion system (2) simplifies to

$$\partial_t(J_\varepsilon^a u_\varepsilon^a(x, t)) - \nabla \cdot (J_\varepsilon^a D_\varepsilon^a \nabla u_\varepsilon^a) = 0, \quad x \in \Omega_\varepsilon^a, \quad t \in S, \quad (5b)$$

$$\begin{aligned} \partial_t u_\varepsilon^w(x, t) - \nabla \cdot (\varepsilon^l D_\varepsilon^w \nabla u_\varepsilon^w) &= -r_\varepsilon f_\varepsilon^r, & x \in \Omega_\varepsilon^w, \quad t \in S, \\ -(J_\varepsilon^a D_\varepsilon^a \nabla u_\varepsilon^a) \cdot \nu_\varepsilon^a &= (\varepsilon^l D_\varepsilon^w \nabla u_\varepsilon^w) \cdot \nu_\varepsilon^w, & x \in \Gamma_\varepsilon, \quad t \in S, \\ -(J_\varepsilon^a D_\varepsilon^a \nabla u_\varepsilon^a) \cdot \nu_\varepsilon^a &= \varepsilon a_\varepsilon (u_\varepsilon^a - u_\varepsilon^w), & x \in \Gamma_\varepsilon, \quad t \in S, \\ -(\varepsilon^l D_\varepsilon^w \nabla u_\varepsilon^w) \cdot \nu_\varepsilon^w &= 0, & x \in \Gamma_\varepsilon^{ws}, \quad t \in S. \end{aligned} \quad (5c)$$

Homogeneous Neumann boundary conditions are prescribed at the exterior boundary $\partial\Omega$ for simplicity as well as initial conditions $(u^a(0), u^w(0)) = (u_0^a, u_0^w)$.

For a function $u = u(x, t) \in L^2(\Omega \times S)$, let $|u(t)|_\Omega$ be the $L^2(\Omega)$ -norm and $|u|_{\Omega, t}$ be the $L^2(\Omega \times (0, t))$ -norm. For the coefficients of the system, we assume that their space variables can each be split into a macroscopic and a microscopic one (often called slow and fast component): For $\alpha \in \{a, w\}$, it is assumed that there exist functions $D^\alpha = D^\alpha(x, y, t)$, $x \in \Omega_\varepsilon^\alpha$, $y \in Z^\alpha$, $t \in S$, bounded from above and from below by numbers D_{\max}^α and D_{\min}^α , respectively, and periodically extended in y , such that $D_\varepsilon^\alpha(x, t) = D^\alpha(x, x/\varepsilon, t)$ and analogously for a_ε and r_ε . We further assume the norm equalities

$$\lim_{\varepsilon \rightarrow 0} |D_\varepsilon^\alpha(t)|_\Omega^2 = |D^\alpha(t)|_{\Omega \times Z^\alpha}^2, \quad \lim_{\varepsilon \rightarrow 0} |r_\varepsilon(t)|_\Omega^2 = |r(t)|_{\Omega \times Z^w}^2, \quad \lim_{\varepsilon \rightarrow 0} \varepsilon |a_\varepsilon(t)|_{\Gamma_\varepsilon}^2 = |a(t)|_{\Omega \times \Gamma}^2 \quad (6)$$

and $r_{\min} \geq 0$ and $D_{\min}^\alpha, a_{\min} > 0$ as well as $\partial_t D_\varepsilon^\alpha, \partial_t a_\varepsilon$ bounded from above. If $C^m > 0$, f_ε^r is also required to be bounded in $L^\infty(S; L^2(\Omega_\varepsilon^w))$.

Since the problem for J_ε^a is decoupled from the reaction–diffusion problem, it can be solved independently beforehand. The reaction–diffusion system is not completely standard owing to the presence of J_ε^a in the parabolic term. For the proof of existence of microsolutions, we make use of an abstract theory for degenerate problems developed in [12]. This enables us to prove the following theorem.

Theorem 2.1 *For given $\varepsilon > 0$, there exists a unique weak solution of problem (5). We have*

$$J_\varepsilon^a(x, t) = 1 - C^m \int_0^t \int_{Z^w} f^r(x, y, s) \, dy \, ds, \quad (7)$$

which satisfies

$$|J_\varepsilon^a(t)|_\Omega^2 \leq C, \quad |\partial_t J_\varepsilon^a|_{\Omega, t}^2 \leq C \quad (8)$$

for a.e. $t \in S$. The functions $(u_\varepsilon^a, u_\varepsilon^w) \in L^2(0, T; W^{1,2}(\Omega_\varepsilon^a)) \times L^2(0, T; W^{1,2}(\Omega_\varepsilon^w))$ satisfy

$$|u_\varepsilon^a(t)|_{\Omega_\varepsilon^a} + |\nabla u_\varepsilon^a|_{\Omega_\varepsilon^a, t} + |u_\varepsilon^w(t)|_{\Omega_\varepsilon^w} + \varepsilon^{1/2} |\nabla u_\varepsilon^w|_{\Omega_\varepsilon^w, t} + \varepsilon^{1/2} |u_\varepsilon^a - u_\varepsilon^w|_{\Gamma_\varepsilon, t} \leq C \quad (9a)$$

for a.e. $t \in S$, where the constant C depends on T and the data but not on ε if $\partial_t J_\varepsilon^a$ is non-negative or bounded independently of ε in $L^\infty(\Omega \times S)$.

3. Identification of the limit problems

For the examination of the convergence of the limit functions as $\varepsilon \rightarrow 0$ and for the identification of the limit problems, two-scale convergence [13,14] and periodic unfolding [11] are used. We only present the identification of the limit problem associated with the volume factor J_ε^a . The complete limit problem of (5) is stated in the next section.

Choosing the test function as $\varphi^J(x, y, t)$ with $\varphi^J \in L^2(\Omega \times Y \times S)$ in the weak form of (5a) and integrating over Y , we obtain

$$\begin{aligned} \int_0^T \int_\Omega \int_Y \mathcal{T}_\varepsilon(\partial_t J_\varepsilon^a)(x, t) \varphi^J(x, y, t) \, dy \, dx \, dt \\ = -C^m \int_0^T \int_\Omega \int_Y \int_{Z^w} \mathcal{T}_\varepsilon(f_\varepsilon^r)(x, z, t) \, dz \, \varphi^J(x, y, t) \, dy \, dx \, dt, \end{aligned} \quad (10)$$

where we have used the fact that J_ε^a is constant in each cell and, therefore, $\partial_t J_\varepsilon^a = \mathcal{T}_\varepsilon(\partial_t J_\varepsilon^a)$ is independent of y . Since the sequences are independent of y , the limit equation as $\varepsilon \rightarrow 0$ is given by

$$\int_0^T \int_\Omega \partial_t J^a(x, t) \int_Y \varphi^J(x, y, t) dy dx dt = -C^m \int_0^T \int_\Omega \int_{Z^w} f^r(x, z, t) dz \int_Y \varphi^J(x, y, t) dy dx dt. \quad (11)$$

Owing to the equivalence of convergences, these are also the two-scale limits and, moreover, it suffices to take φ^J independent of y . Therefore,

$$\int_0^T \int_\Omega \partial_t J^a(x, t) \varphi^J(x, t) dx dt = -C^m \int_0^T \int_\Omega \int_{Z^w} f^r(x, y, t) dy \varphi^J(x, t) dx dt \quad (12)$$

for all $\varphi^J \in L^2(\Omega \times S)$.

4. The macroscopic limit problem

We state the macroscopic limit problem of (5) as $\varepsilon \rightarrow 0$, where the limit functions of J_ε^a , u_ε^a and u_ε^w are denoted by J^a , u^a and u^w , respectively. We define a generic cell problem: Let the functions ζ_j^α , $j = 1, \dots, N$, be the Y -periodic solution of the cell problem

$$-\nabla_y \cdot (A(x, y, t)(\nabla_y \zeta_j^\alpha(x, y, t) + e_j)) = 0, \quad y \in Z^\alpha, \quad x \in \Omega, \quad t \in S, \quad (13)$$

with homogeneous Neumann conditions at all interior boundaries, the weak form of which is given by

$$(A(x, \cdot, t)(\nabla_y \zeta_j^\alpha(x, \cdot, t) + e_j) | \nabla_y \phi)_{Z^\alpha} = 0 \quad (14)$$

for all Y -periodic test functions ϕ . The vector e_j is the j th unit vector in N -dimensional Euclidean space. Writing δ_{ij} for the Kronecker delta, the ζ_j^α are used to define the tensor $P^\alpha = [p_{ij}^\alpha]_{ij}$ as

$$p_{ij}^\alpha(x, t) = \int_{Z^a} A(x, y, t)(\delta_{ij} + \partial_{y_i} \zeta_j^\alpha(x, y, t)) dy. \quad (15)$$

We will specify A and α to our needs below.

The limit problem for J^a is given by

$$\partial_t J^a(x, t) = -C^m \int_{Z^w} f^r dy, \quad x \in \Omega, \quad t \in S, \quad (16)$$

with initial condition $J^a(0) = 1$. The limit equation for u^a states

$$\partial_t (J^a u^a(x, t)) - \nabla \cdot (P^a \nabla u^a) = - \int_\Gamma a(u^a - u^w) d\sigma_y, \quad x \in \Omega, \quad t \in S, \quad (17)$$

with homogeneous Neumann conditions at $\partial\Omega$ and initial condition $u^a(0) = u_0^a$, where P^a is the tensor defined by (15) with $A = J^a D^a$ and $\alpha = a$.

The limit equation for u^w depends on the particular choice of the scaling exponent l . If $l < 2$, the limit equation is given by

$$\partial_t u^w(x, t) - \delta_{0l} \nabla \cdot (P^w \nabla u^w) = - \int_{Z^w} r f^r dy + \int_\Gamma a(u^a - u^w) d\sigma_y, \quad x \in \Omega, \quad t \in S, \quad (18)$$

where P^w is defined by (15) with $A = D^w$ and $\alpha = w$. If $l = 2$, the limit problem is given by

$$\partial_t u^w(x, y, t) - \nabla_y \cdot (D^w \nabla_y u^w) = -r f^r, \quad x \in \Omega, \quad y \in Z^w, \quad t \in S, \quad (19)$$

$$-D^w \nabla_y u^w \cdot \nu^w = a(u^a - u^w), \quad x \in \Omega, \quad y \in \Gamma, \quad t \in S, \quad (20)$$

$$-D^w \nabla_y u^w \cdot \nu^w = 0, \quad x \in \Omega, \quad y \in \Gamma^{ws}, \quad t \in S. \quad (21)$$

For all choices of l , homogeneous Neumann conditions apply at the exterior boundaries and $u^w(0) = u_0^w$.

Thus, the limit problem consists of one ordinary differential equation for the macroscopic pore-air volume factor $J^a(x, t)$ coupled to a reaction–diffusion problem for the concentrations u^a and u^w , where we have $u^a = u^a(x, t)$ as well as $u^w = u^w(x, t)$ if $l < 2$ and $u^w = u^w(x, y, t)$ if $l = 2$.

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